

Making Meaning out of Learning the Rise of Applied Learning

Dr. Chris Slatter – Senior Teacher (Chemistry)

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- Redesigning Pedagogy International Conference
2nd – 4th June 2015



APPLIED LEARNING AT SST

- This presentation can be found online at...



www.scientist.sg

School of Science and Technology, Singapore

- The School of Science and Technology, Singapore (SST) is a specialised independent school with a cohort of ca. 800 students. Entry to SST is through Direct School Admission.
- SST offers a 4-year programme that focuses on science and technology delivered through an applied learning framework.
- Apart from a sound broad based academic curriculum, SST also offers four *applied subjects*, namely Biotechnology, Fundamentals of Electronics, Media Studies and Design Studies.
 - As a “Future School”, SST leverages Information Communication Technology (ICT) pervasively to enhance the learning environment of its students.



**SCHOOL OF SCIENCE AND
TECHNOLOGY, SINGAPORE**

LEARNING IN A WHOLE NEW WAY

APPLIED LEARNING AT THE SCHOOL OF SCIENCE AND TECHNOLOGY, SINGAPORE

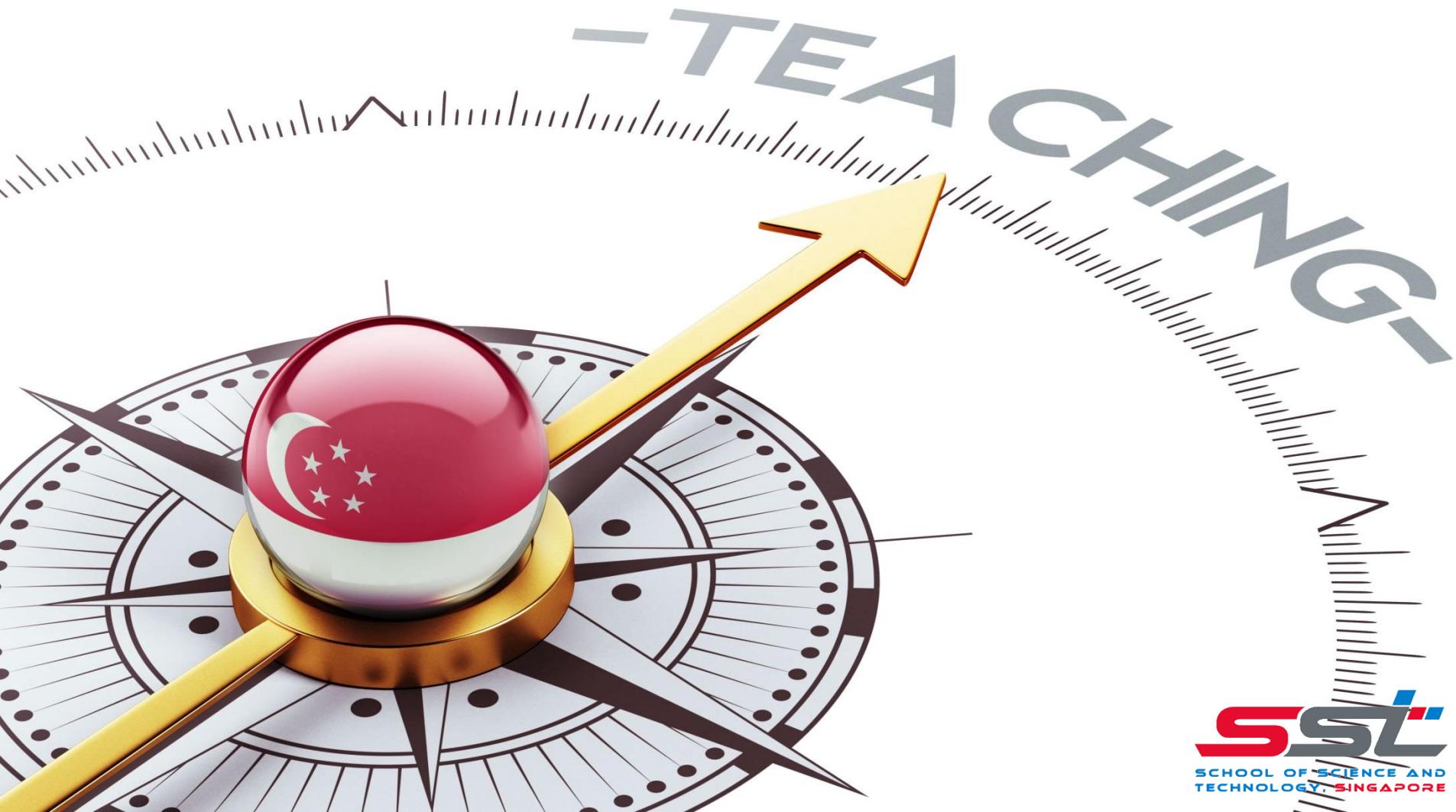
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- Singapore has a world class education system that it must continue to invest in.



- What direction should teaching and learning take in the future?



‘Abundance’ mentality that recognises multiple pathways to success needed, says Education Minister

“While changes in the economy have created jobs requiring specific skills and multiplied pathways, our thinking about education and pathways have not kept up with these changes,” [Mr. Heng said].

“One way the Education Ministry has tried to change such thinking is through the *Applied Learning Programme* and Learning for Life Programme, and to create opportunities in every school”.

“Parents are beginning to see how each school can offer something interesting and special,” [Mr. Heng said].

- Laura Philomin, Tuition culture has to go, say MPs, *Today*, Saturday 7th March 2015, page 9, MediaCorp Press Ltd.

APPLIED LEARNING AT SST



Essential Question:

How do we engage, motivate and prepare the students of today...

...to become the scientists and engineers of tomorrow?

- How should we teach?
- How should they learn?

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“Every kid starts out as a natural born scientist, and then we beat it out of them. A few trickle through the system with their wonder and enthusiasm for science intact.”

Carl Sagan, 1934 - 1996



APPLIED LEARNING AT SST



Video = 10.8 seconds

- How did you learn to ride a bicycle?

APPLIED LEARNING AT SST

How did you learn to ride a bicycle?

- Read a **book** about how to ride a bike?

- Watch a **video** about how to ride a bike?



- Attend a **lecture** about how to ride a bike?

- Take a **multiple-choice test** about how to ride a bike?

APPLIED LEARNING AT SST

- Or did you learn to ride a bicycle like this?

APPLIED LEARNING AT SST

- Or did you learn to ride a bicycle like this?

APPLIED LEARNING AT SST

- Or did you learn to ride a bicycle like this?

APPLIED LEARNING AT SST



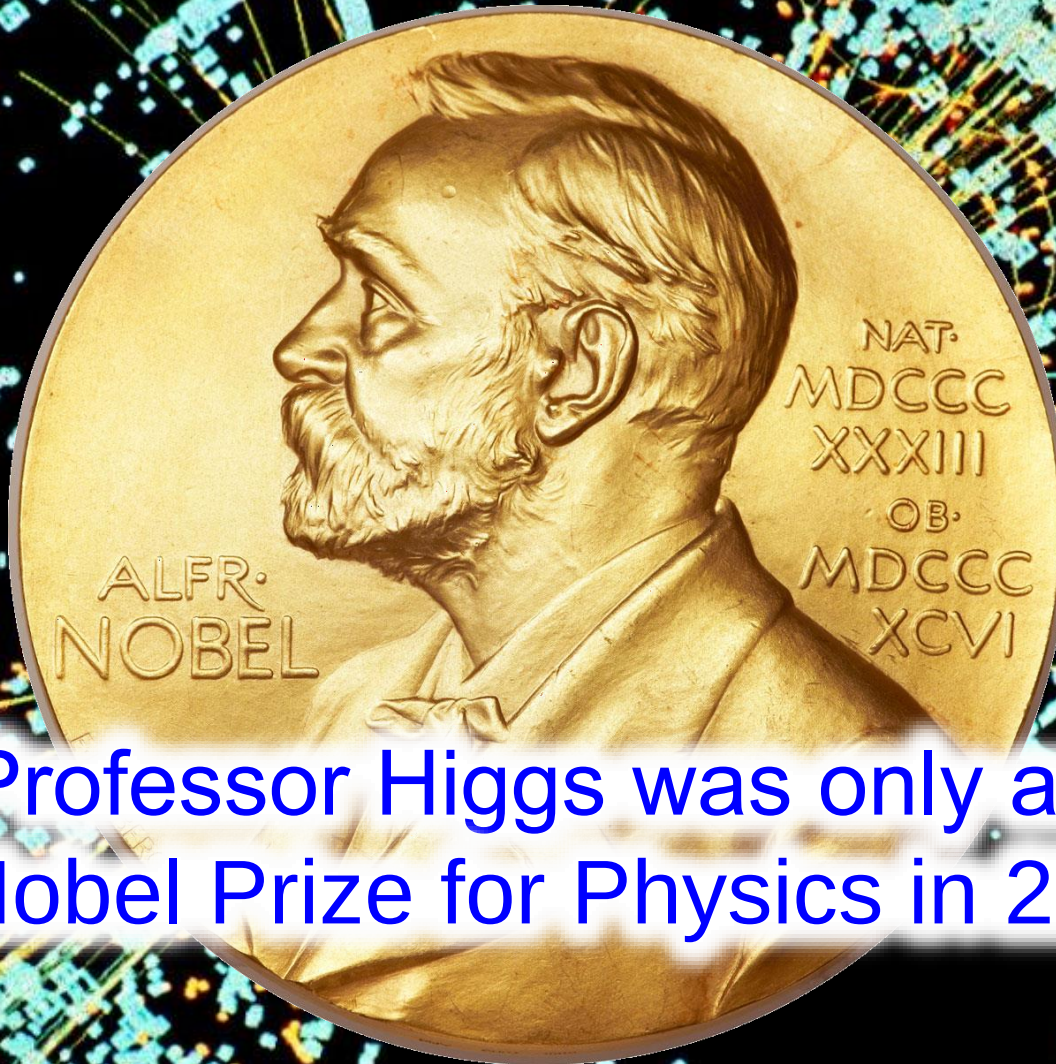
Video = 13.5 seconds

“Science” is a verb.

APPLIED LEARNING AT SST

Professor Peter Higgs (left) proposed the existence of the Higgs boson in 1964...

APPLIED LEARNING AT SST



...but Professor Higgs was only awarded the Nobel Prize for Physics in 2013...

APPLIED LEARNING AT SST

...because the Higgs boson was only detected by the Compact Muon Solenoid at CERN on 4th July 2012.

The Applied Learning approach guides the SST teaching and learning process so that the curriculum is delivered within a relevant real-life context to ensure that learning experiences are coherent, challenging and aimed at developing the full range of a student's capabilities and talents.

APPLIED LEARNING AT SST

The **Applied Learning** approach guides the SST teaching and learning process so that the curriculum is delivered within a relevant **real-life context** to ensure that learning experiences are **coherent, challenging** and aimed at developing the full range of a **student's capabilities and talents**.

APPLIED LEARNING IN A NUTSHELL



APPLIED LEARNING IN A NUTSHELL

- Applied learning is the application of learning.
- Students demonstrate understanding by transferring their knowledge to solve novel problems framed in a real world context.
- Students use critical and creative thinking skills while they collaborate on solving authentic problems.
 - Guiding students to “**think like a disciplinarian**” (think like a chemist).

APPLIED LEARNING AT SST

Why do I
need to
know this?

APPLIED LEARNING AT SST



Why do I
need to
know this?

It's tested
on the
exam.

APPLIED LEARNING AT SST

Let's try that again...

APPLIED LEARNING AT SST



Why do I
need to
know this?

APPLIED LEARNING AT SST



Why do I
need to
know this?

It's interesting
and has a real
world application.

APPLIED LEARNING LESSON PLANNING

- What do I want my students to understand?
- Why is it important for them to understand this?
- What authentic problems can they solve with the cognitive skills that they have learnt?
- What authentic problems can they solve with the procedural skills that they have learnt?
- How will they demonstrate their understanding (beyond a grade on a pen-and-paper test)?

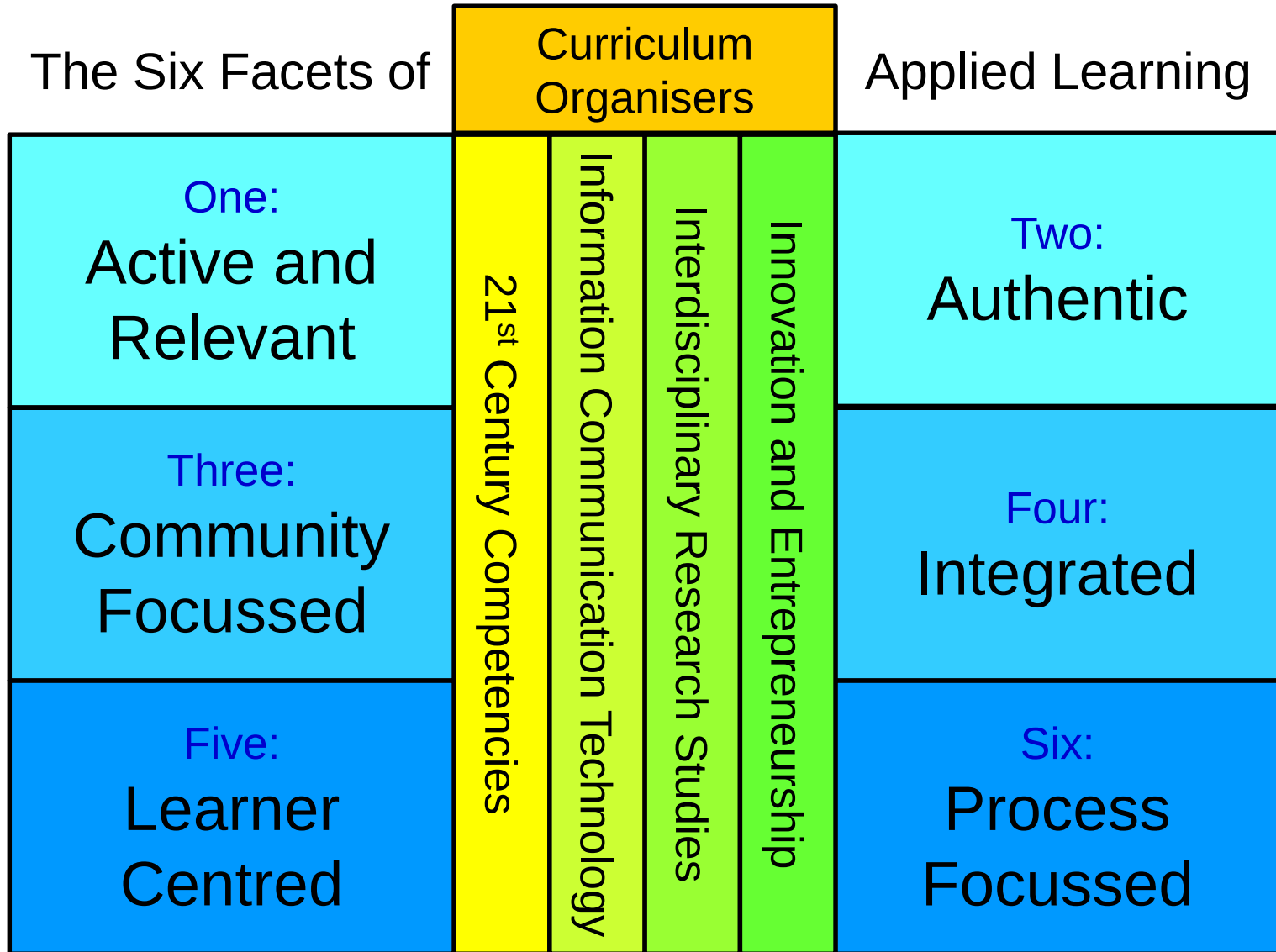
APPLIED LEARNING FRAMEWORK

The Six Facets of

Curriculum Organisers			
Innovation and Entrepreneurship			
Interdisciplinary Research Studies			
Information Communication Technology			
21 st Century Competencies			

Applied Learning

APPLIED LEARNING FRAMEWORK



ACTIVE AND RELEVANT

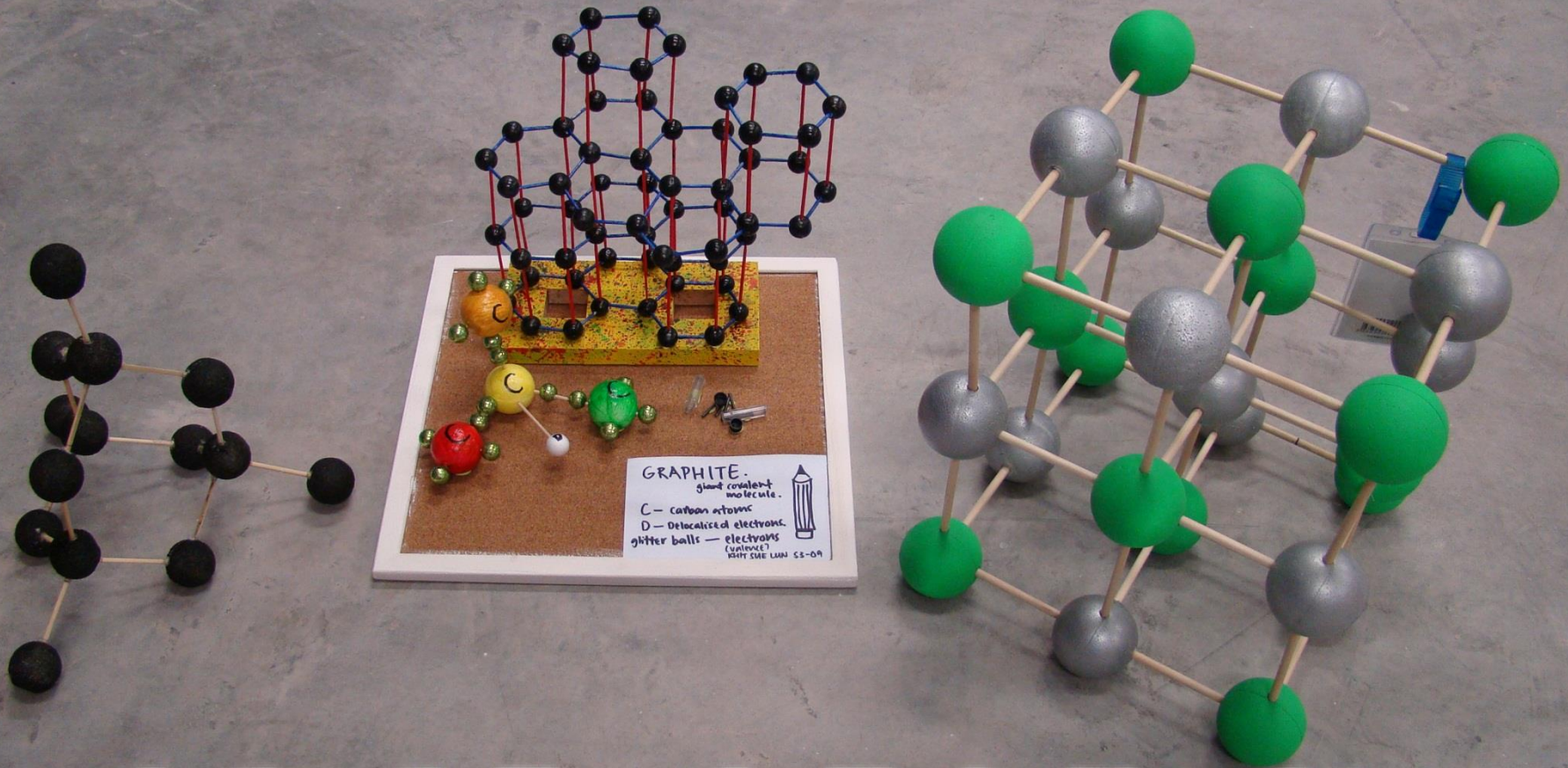
- Engaging the learner as co-constructor of knowledge and meaning.
- Providing challenge through rigorous disciplinary content.
- Learning by doing and experiencing.

APPLIED LEARNING AT SST



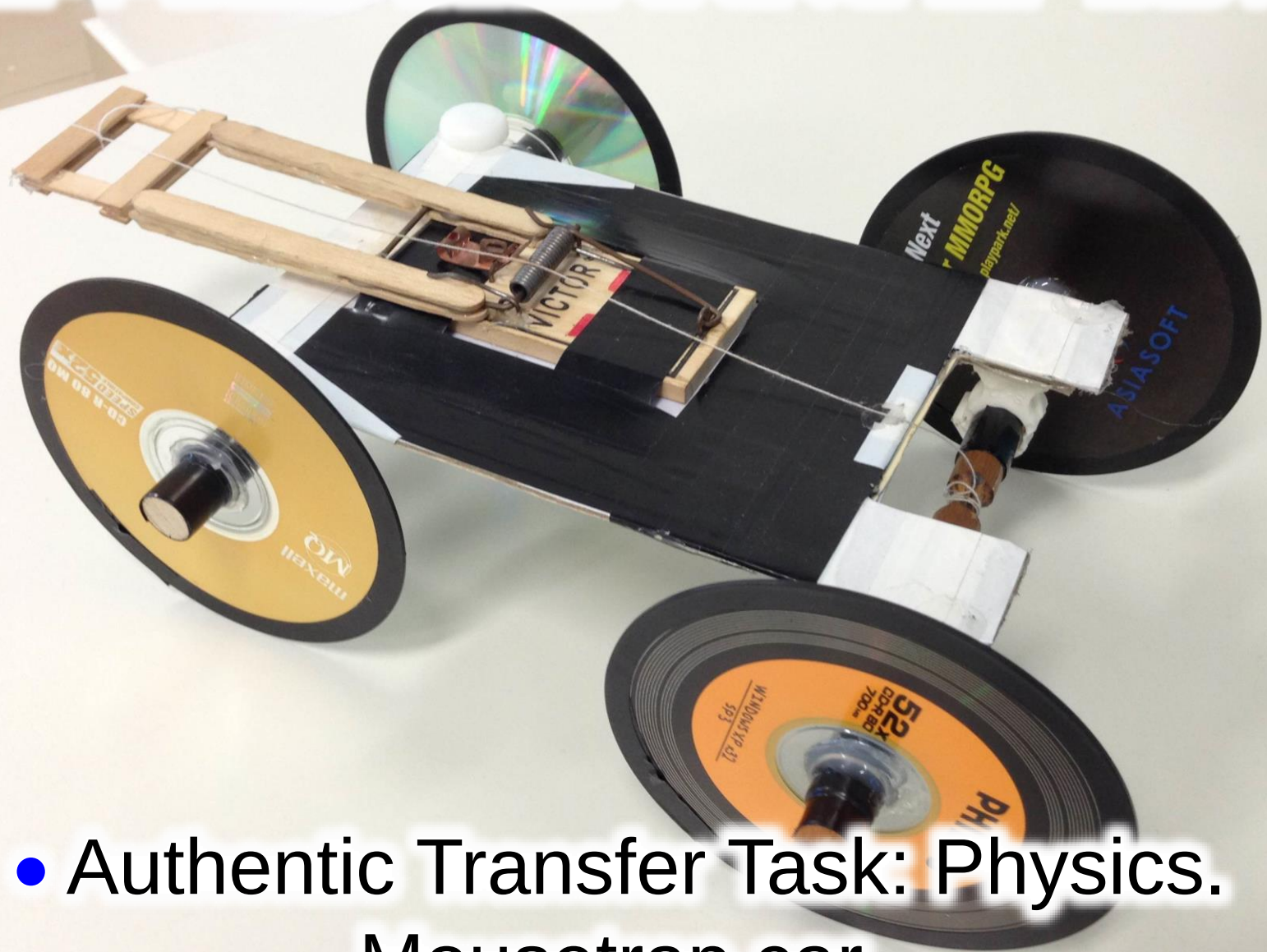
- Authentic Transfer Task: Chemistry.
Molecular models.

APPLIED LEARNING AT SST



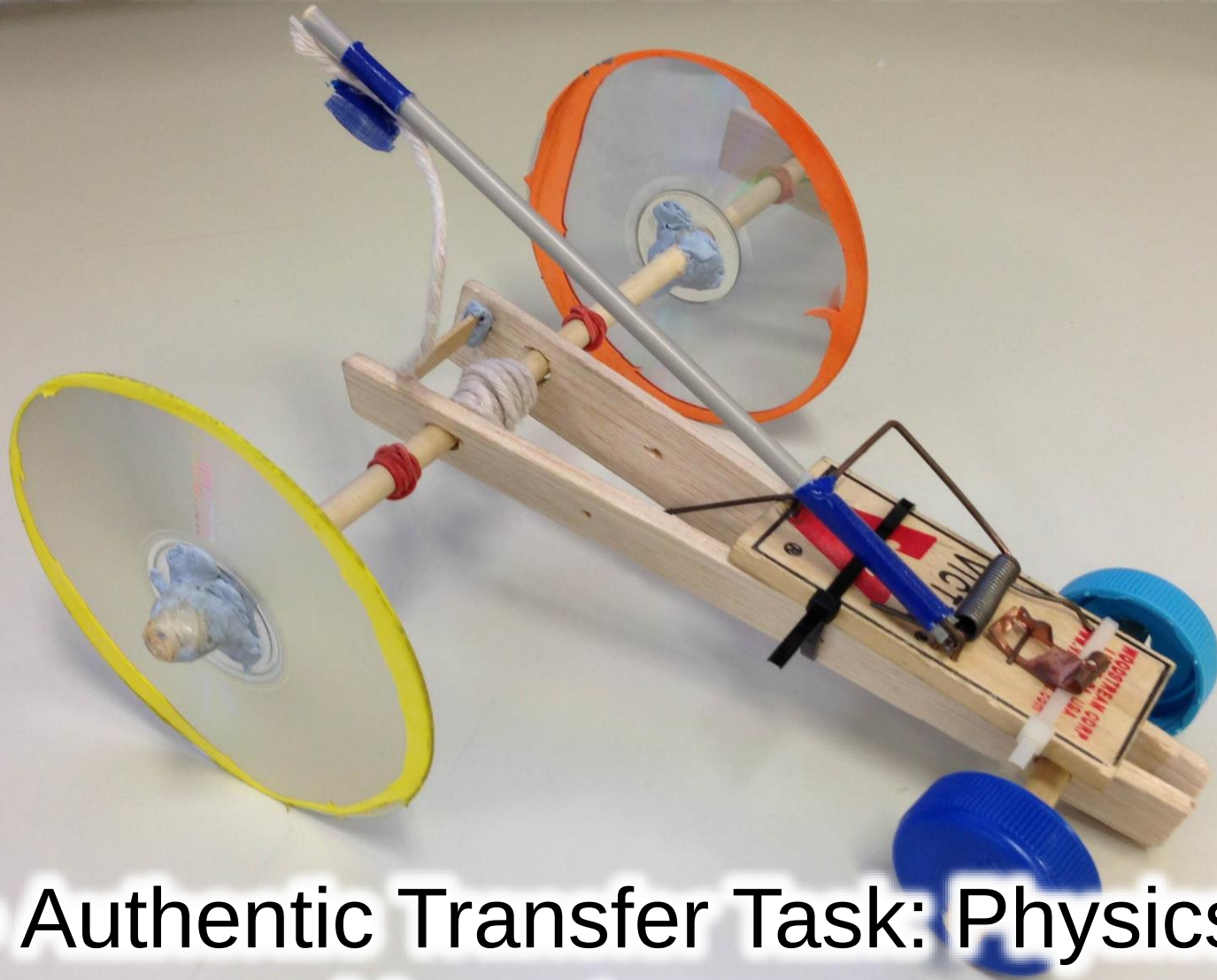
- Authentic Transfer Task: Chemistry.
Molecular models.

APPLIED LEARNING AT SST



- Authentic Transfer Task: Physics.
Mousetrap car.

APPLIED LEARNING AT SST



- Authentic Transfer Task: Physics.
Mousetrap car.

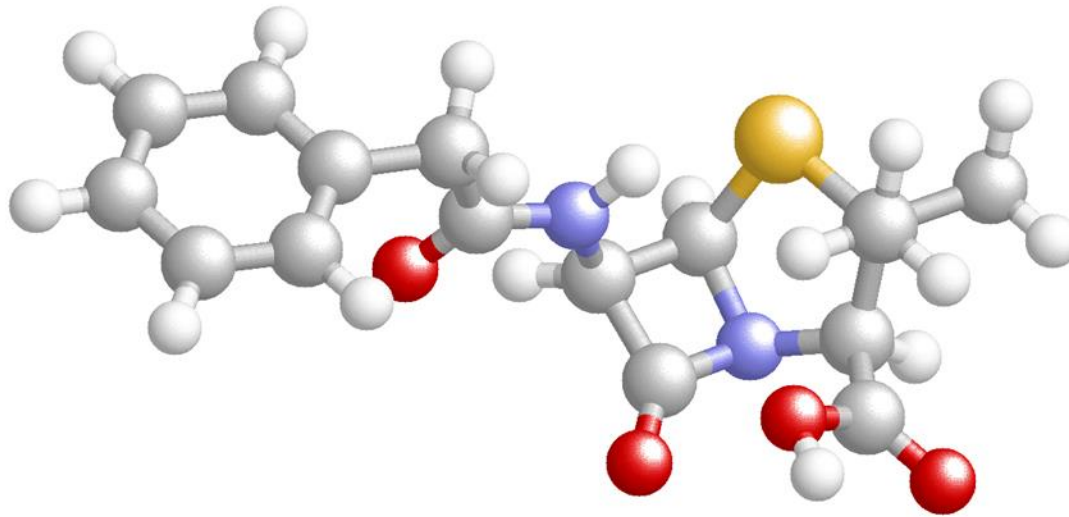


- Authentic Transfer Task: Biology.

- Singapore produced enough food waste last year to fill *600 Olympic size swimming pools*.
- In the recent years, there are many enthusiasts who promote recycling of kitchen waste without affecting the environment. One of the common terms is “*Garbage Enzymes*”.
- As ambassadors for the environment, the students have been tasked to research on how “*Garbage Enzymes*” work, and educate their peers on their findings.

AUTHENTIC

- Contextualising learning with tasks and resources drawn from, and applicable to, the real world.
- Developing multi-modal ways of learning.
 - Learning synchronously and asynchronously.



Penicillin

- Students study the history of antibiotics and then go on to make and test a semi-synthetic penicillin.



SST
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Chem!stry

Name: _____ ()

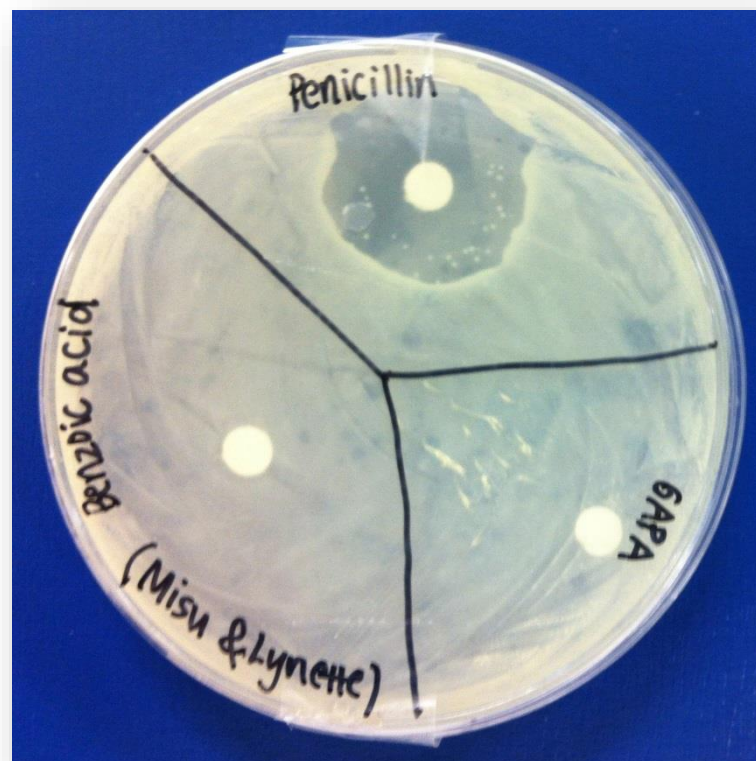
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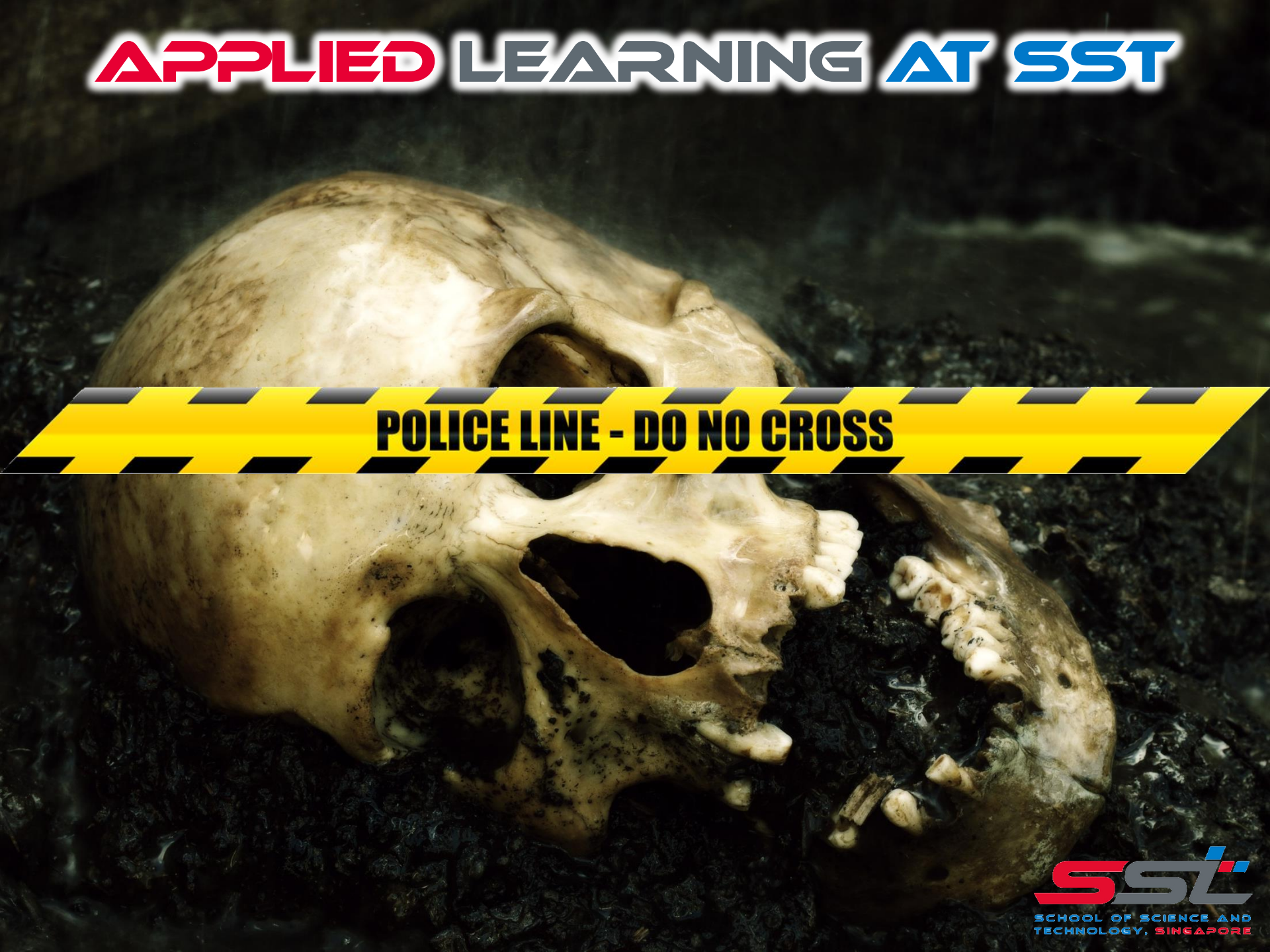
Making and Testing a Penicillin - Basic

Apparatus: 100 cm³ stoppered bottle or conical flask 2 × 25 cm³ measuring cylinders 10 cm³ measuring cylinder Test pipettes or droppers Test tubes 2 × 100 cm³ beakers Glass rod or magnetic stirrer Universal indicator paper or pH meter 50 cm³ separating funnel 5 – 7 mm cork borer Adhesive tape Protective gloves Protective goggles Reagents: 6-aminopenicillanic acid 5.0 cm³ of 1.0 mol dm⁻³ aqueous sodium hydroxide 0.5 cm³ benzoyl chloride 5.0 cm³ propanone 15.0 cm³ ethyl ethanoate 10 cm³ of 1.0 mol dm⁻³ hydrochloric acid 25 cm³ saturated sodium hydrogencarbonate solution 4 agar plates impregnated with <i>Bacillus subtilis</i> Ethanol (for sterilising apparatus) Beaker of disinfectant (for sterilising apparatus and work surfaces) Control solution of 6-aminopenicillanic acid Control solution of sodium benzoate	Care: 6-aminopenicillanic acid can act as a sensitiser by inhalation or skin contact. Wear protective gloves and do not inhale the dust. Care: You should not perform this activity if you are allergic to penicillin. Care: Benzoyl chloride is corrosive and a lachrymatory agent (it is a severe eye irritant) and must be used in a fume cupboard. It produces fumes of hydrogen chloride gas when exposed to moist air. Wear gloves and eye protection when handling this reagent. Care: Propanone, ethyl ethanoate and ethanol are highly flammable liquids. Keep bottles stoppered when not in use and keep these reagents well away from naked flames. Care: Consult your teacher before handling the bacterial culture. Wear gloves at all times. Cover any skin cuts with an effective waterproof dressing and wash your hands thoroughly at the end of the session. Report any spillages immediately. Any material that has come into contact with the bacterial culture must be sterilised before disposal.	 Irritant  Corrosive  Highly Flammable  Biohazard
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1



APPLIED LEARNING AT SST



POLICE LINE - DO NO CROSS

Crime scene investigation for qualitative analysis.

POLICE LINE - DO NO CROSS

- Students analyse a compound found at a crime scene to identify which anions and cations it contains.
- Students then submit a forensic science report to the police.

APPLIED LEARNING AT SST



Map data © Google 2015, © Urban Redevelopment Authority Imagery 2015.

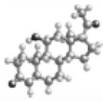
Google

APPLIED LEARNING AT SST

- The body of a young man has been found on Marina Bay Golf Course.
- The cause of death is drowning.
- You are provided with a sample of water from the victim's lungs.
- You are also provided with water samples from Marina Bay, Geylang River (near a construction site) and the sea.
- By qualitative analysis, identify the anions and cations in all four samples of water and hence determine where the victim drowned.

APPLIED LEARNING AT SST





Question 1:

A seven year-old
suffering a compo
morphine sulphat
sulphate oral solu



Question 2:

A 70.0 kg adult m
production of neu
administer to the



Question 3:

An 80.0 kg adult male requires 120 units kg^{-1} of Epopen to be administered for anaemia, secondary to chemotherapy. What volume of Epopen will you administer to the patient by intravenous injection?



Question 4:

A 42.0 kg female requires 10 mg kg^{-1} of Digoxin to be administered every eight hours to combat heart failure. What volume of Digoxin will you administer to the patient orally?



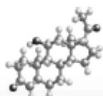
Question 5:

An 81.0 kg male requires 5 mg kg^{-1} Clindamycin phosphate to be administered every day, equally divided into three doses, to combat a respiratory tract infection. What volume of Clindamycin will you administer to the patient by intravenous infusion each time?



- Students apply their knowledge of mole calculations to drug calculations and determining the concentration of electrolytes in a patients' blood.

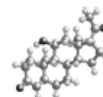
APPLIED LEARNING AT SST



Name: _____ ()

Class: _____

Date: ____/____/____



Name: _____ ()

Class: _____

Date: ____/____/____

Experiment to determine the Mass of Iron(II) Sulphate in Iron Tablets

Introduction

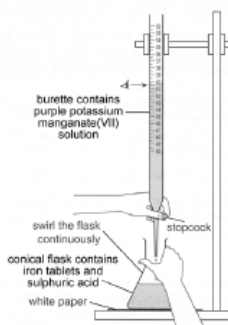
Iron tablets are prescribed for anaemia. A single iron tablet typically contains 200 mg (0.2 g) of Iron(II) sulphate (formula FeSO_4). This can be checked by titration.

Health and Safety

Do not allow any of the solutions to come into contact with your skin. If a solution does come into contact with your skin, wash the affected area thoroughly with cold water.

Method

- 1) Accurately weigh three iron tablets.
- 2) Grind up the tablets using a pestle and mortar and transfer the resulting powder into a 100 cm^3 volumetric flask.
- 3) Make up to volume using 0.1 mol dm^{-3} sulphuric acid.
- 4) Stopper the flask and shake well. Note that the outer coating of the tablets will probably not dissolve. This will not affect your results.
- 5) Pipette 25.00 cm^3 of the iron tablet/sulphuric acid solution into a 250 cm^3 conical flask.
- 6) Fill a burette with 0.02 mol dm^{-3} potassium manganate(VII) (formula KMnO_4) solution and record the initial burette reading.
- 7) Run the solution of potassium manganate(VII) from the burette into the conical flask containing the iron tablet/sulphuric acid. During the reaction the purple colour of the potassium manganate(VII) will fade to colourless. Stop the titration and record the final burette reading when there is a permanent pale pink colour in the conical flask.
- 8) Repeat the titration until you obtain results that are $\pm 0.10 \text{ cm}^3$.



- Students determine the amount of aspirin in aspirin tablets and the amount of iron in iron(II) sulfate tablets by titration.

APPLIED LEARNING AT SST

As a Chemist,
how can you
help this child
have access
to clean
drinking
water?

APPLIED LEARNING AT SST

Imagine that the following impurities are present in the water:

- a) Copper(II) sulfate
- b) Sand (silica)
- c) Iron filings
- d) Ammonium Chloride
- e) Oil

How would you remove them?

COMMUNITY FOCUSED

- Building a borderless community of learners.
- Inculcating respect for diverse ideas.
- Developing stewardship of issues in the community and world at large.

APPLIED LEARNING AT SST

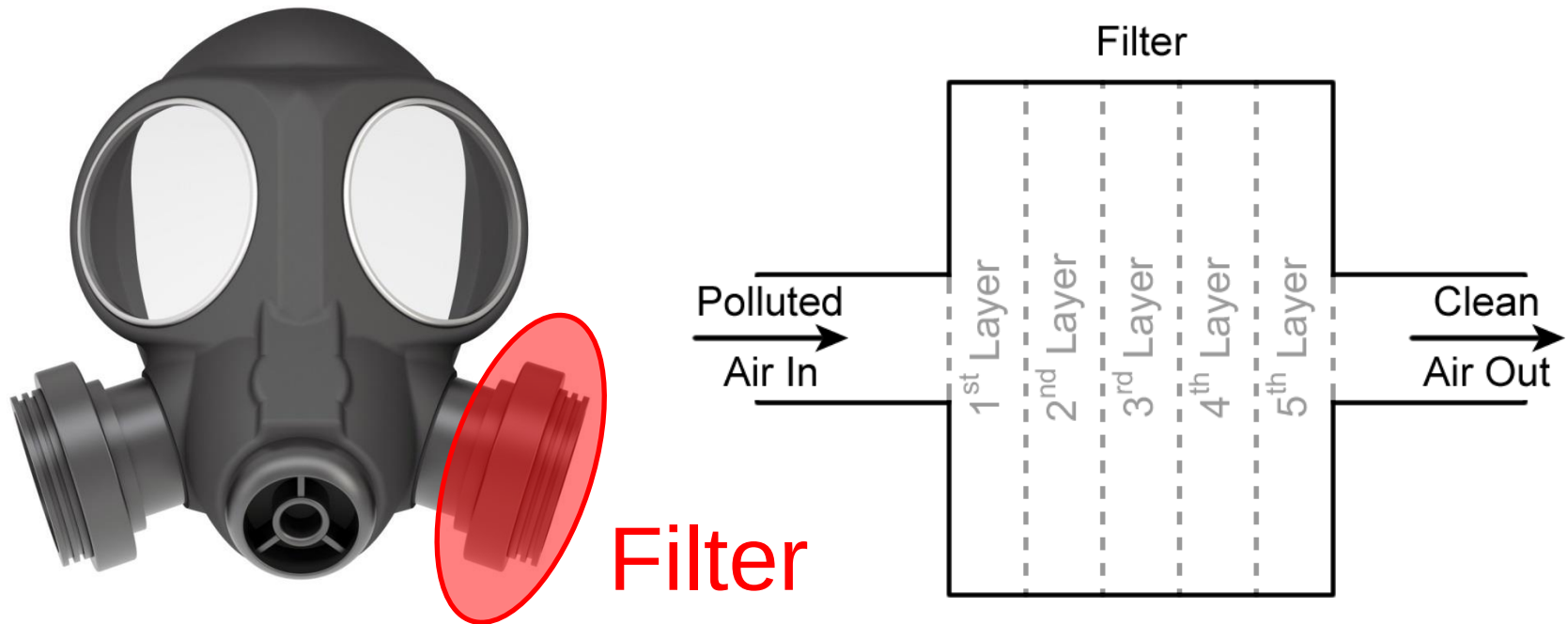
Atmosphere – Community Focus



Atmosphere – Community Focus

- Which atmospheric pollutants are contained within the yearly *haze* that Singapore experiences due to “slash-and-burn” cultivation in neighbouring countries?
- To what extent are paper facemasks effective at removing atmospheric pollutants contained within the *haze*?

Atmosphere – Community Focus



- Students design a gas mask based upon what they have learnt about atmospheric pollution.

Separation and Purification

- **Essential Question:**
Why it is important for these different substances to be free from contamination?



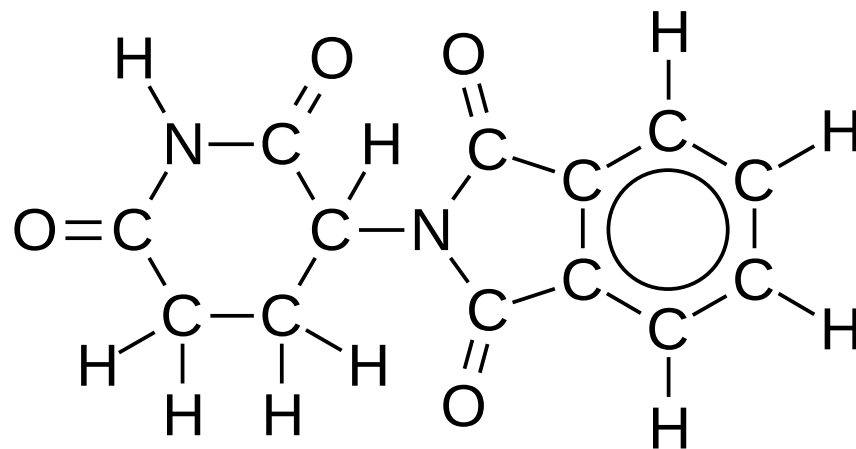
Separation and Purification

To illustrate why it is important for food and drugs to be free from contamination, use your learning device to search for information on the drug *thalidomide*.

- What condition was thalidomide used to treat?
- What serious health problems arose from the use of thalidomide? Why?
- During the manufacture of the drug, what could have been done to avoid these serious health problems?



Separation and Purification



Thalidomide was used by pregnant women in the 1950s and 1960s to combat the effects of morning sickness. The thalidomide molecule exists in two different forms called *isomers*. One of these isomers was successful in reducing the effects of morning sickness, but the other isomer was found to be *teratogenic* (causes deformities in a developing foetus). Because the two isomers *were not separated* from each other, many of the women who took the drug thalidomide gave birth to children with serious physical deformities.

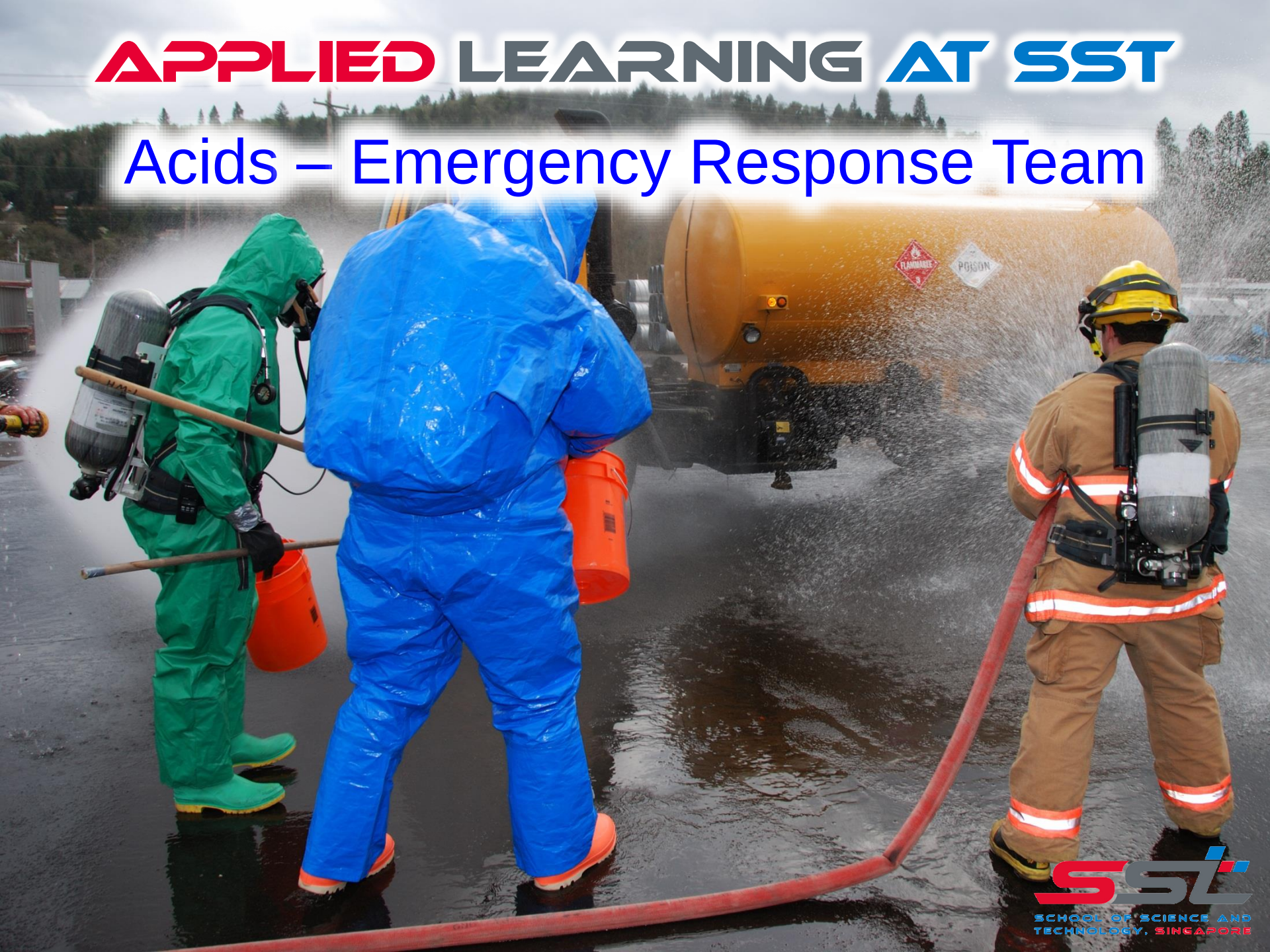
APPLIED LEARNING AT SST

INTEGRATED

- Intra and inter-disciplinary learning and problem solving.
- Teaching for enduring understanding and transference.
- Strengthening values, leadership and citizenship.

APPLIED LEARNING AT SST

Acids – Emergency Response Team



APPLIED LEARNING AT SST

Acids – Emergency Response Team



Acids – Emergency Response Team

- A road tanker carrying 30 000 dm³ of an unknown solution has been involved in an accident with another vehicle.
- The incident has taken place on a road directly adjacent to a large reservoir.
- The drivers of both vehicles have been stabilised at the scene and transported to hospital for further treatment.
- The accident has resulted in the road tanker overturning. It is currently positioned very close to the edge of the reservoir, and is leaking an unknown solution from a hole in its side.

APPLIED LEARNING AT SST

Acids – Emergency Response Team



Acids – Emergency Response Team

- You are a highly qualified specialist in charge of a small team of people trained to deal with the large scale spillage of chemicals.
- You arrive at the scene of the accident 10 minutes after the first police car and ambulance.

POLICE LINE - DO NO CROSS

- The area has already been cordoned-off, and members of the general public are held back at a safe distance.
- You immediately notice that some of the unknown solution is leaking from the road taker into the reservoir.

APPLIED LEARNING AT SST



Acids – Emergency Response Team

- Your immediate task is to identify the unknown solution that is leaking from the road tanker.
 - Secondly, you should identify the hazards that the unknown solution poses to your team and the environment around the road tanker.
 - Thirdly, based upon what you have learned about the unknown solution, you should propose a rapid and effective way of containing and neutralising the chemical spillage.
 - Finally, you should submit a detailed scientific report of your findings to the chief scientific officer:
christopher_slatter@sst.edu.sg

APPLIED LEARNING **AT SST**

Acids – Emergency Response Team



Acids – Emergency Response Team

- The meteorological office reports that a storm, bringing heavy rain, will move into your area from the South-East within the next 50 minutes.
- Consequently, you have been given a maximum of 40 minutes to complete this emergency exercise.

APPLIED LEARNING AT SST

LEARNER CENTRED

- Honouring student talents and interests.
 - Differentiating learning experiences.
- Developing dispositions for learning while learning.

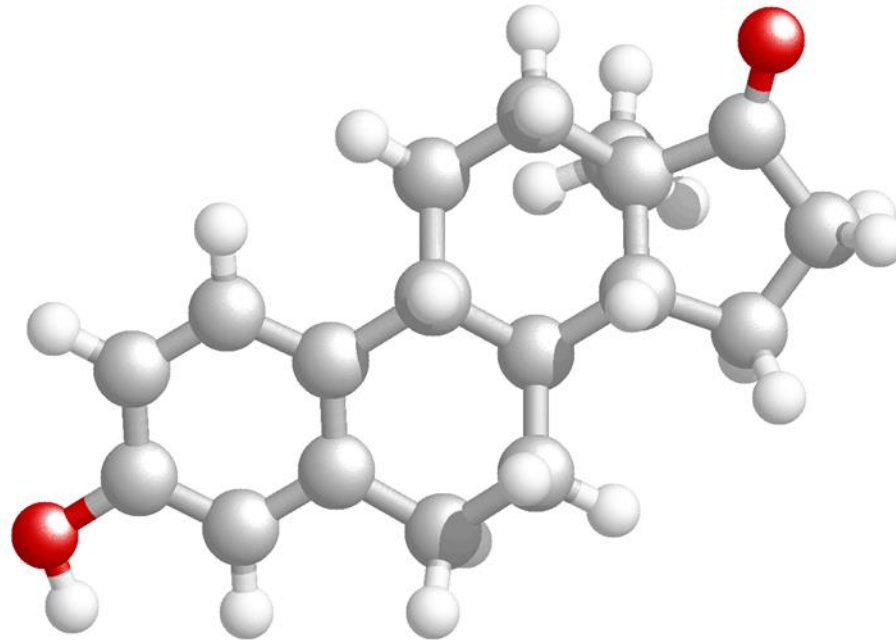
P A R E N T A L

ADVISORY

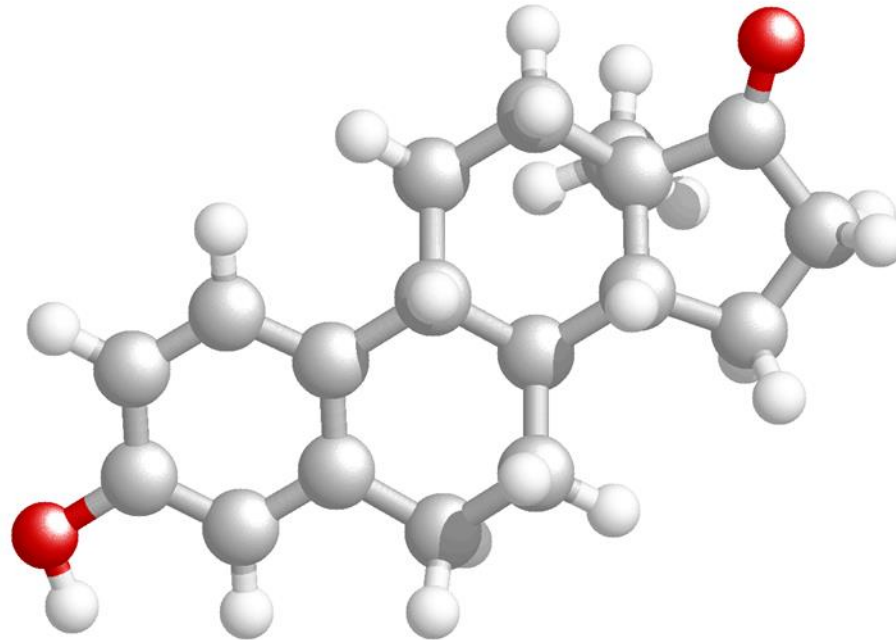
EXPLICIT CONTENT



The Chemistry of Love



♥ *Hormones* are chemical messengers that travel throughout the body in the circulatory system.



♥ Compared to signals that are sent through the body via the central nervous system, signals sent using hormones have a *slower* response time but have a *longer lasting effect*.



Photographs of identical twin sisters with different oestrogen levels.

- ♥ Research carried out at the University of St. Andrews suggests that women with high levels of the sex hormone *oestrogen* have prettier faces. The findings make evolutionary sense because it means that men are attracted to the most fertile women. Oestrogen levels during puberty can impact on appearance by affecting bone growth and skin texture.



Photographs of identical twin sisters with different oestrogen levels.

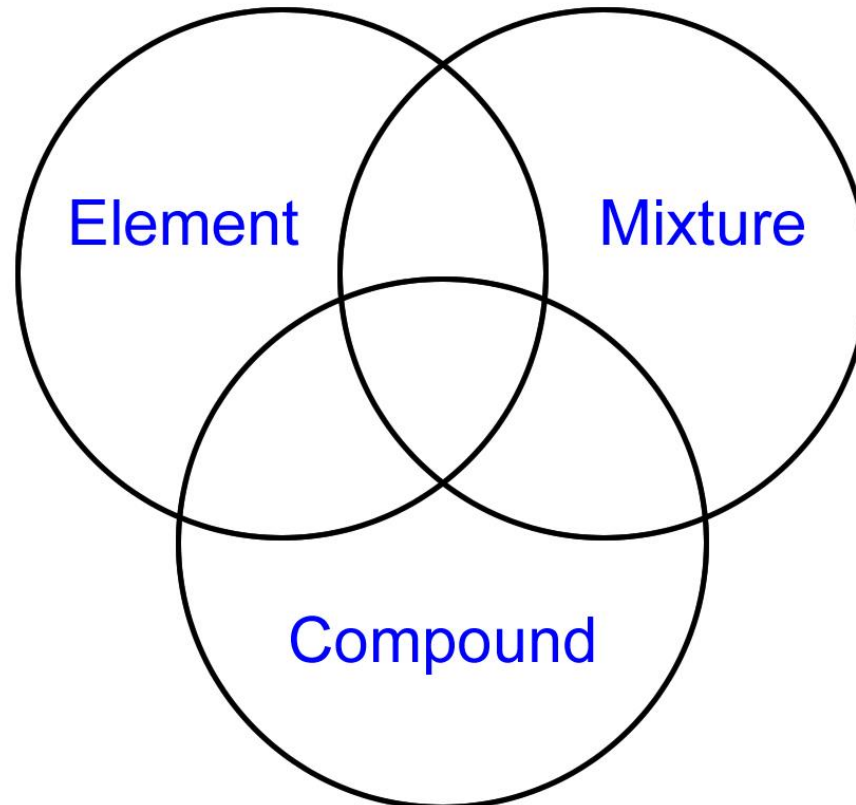
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PROCESS FOCUSED

- Fostering higher-order-thinking through critical thinking models and creative problem solving strategies.
- Uncovering the heart of disciplines through inquiry.
- Teaching disciplines as a way of thinking and doing.

Higher Order Thinking Skills

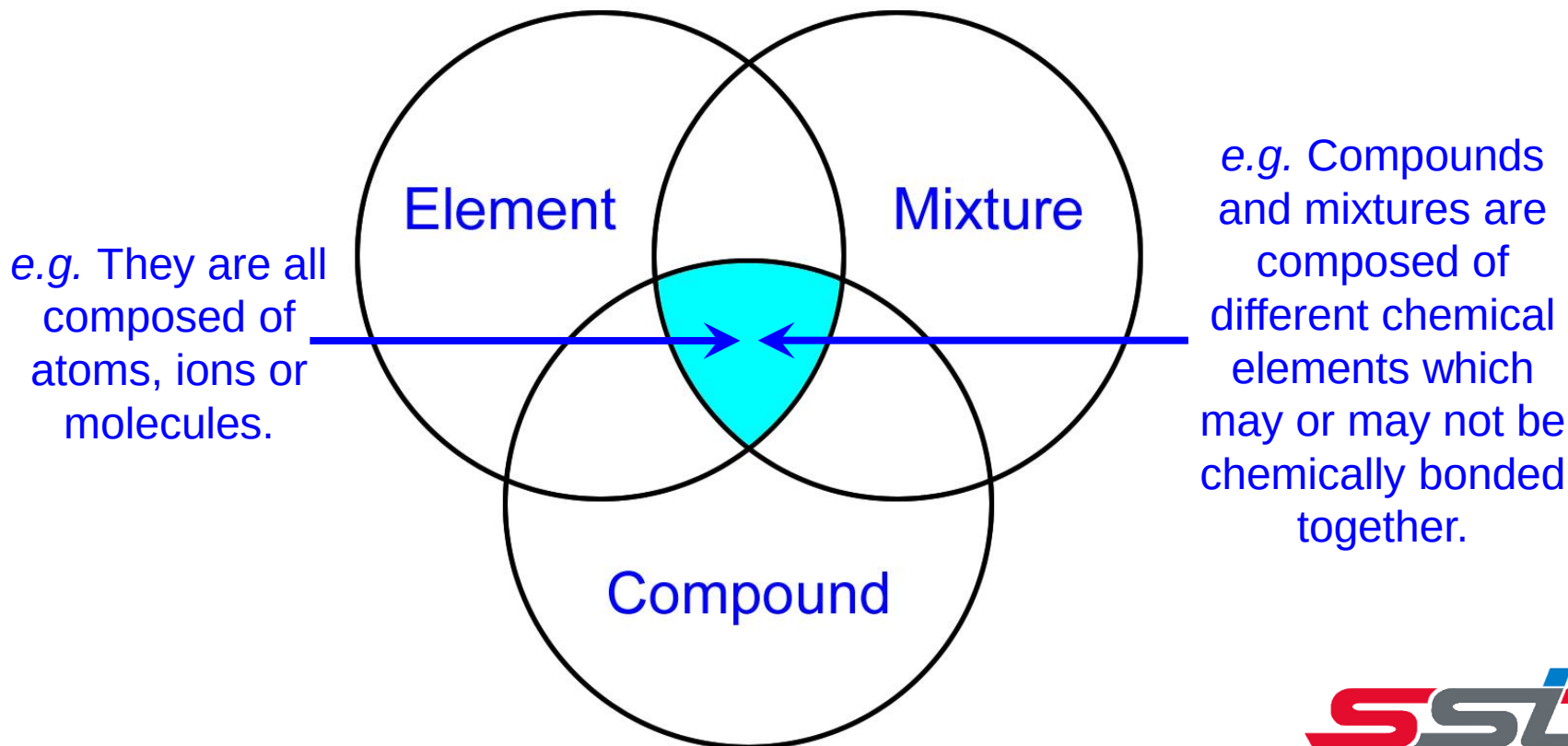
- Roberts, J. L., & Inman, T. F. (2007). *Strategies for differentiating instruction*. Waco, Texas: Prufrock Press, Inc.



Higher Order Thinking Skills

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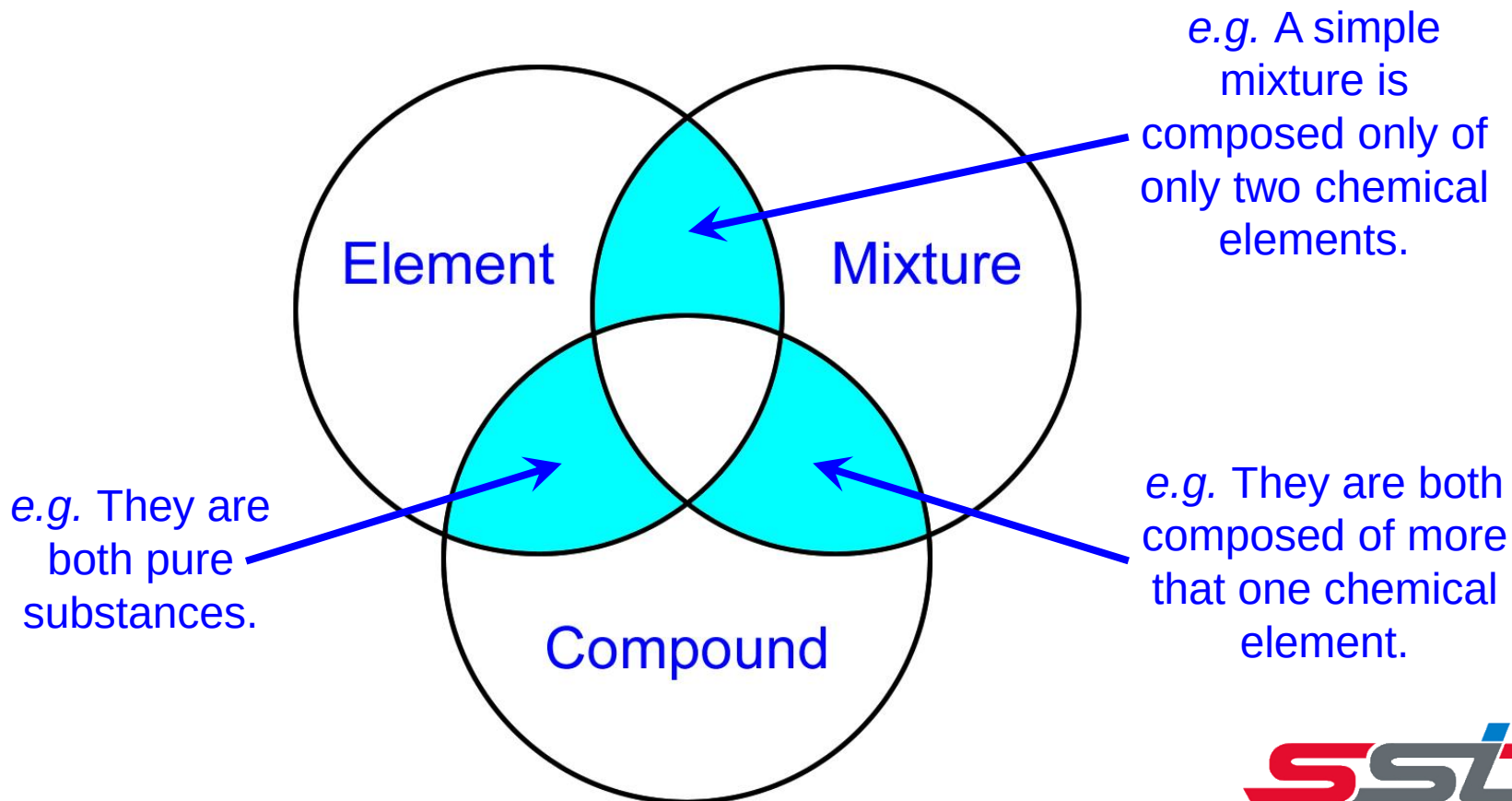
In what ways are elements, compounds and mixtures *all similar* to each other?



Higher Order Thinking Skills

- Roberts, J. L., & Inman, T. F. (2007). *Strategies for differentiating instruction*. Waco, Texas: Prufrock Press, Inc.

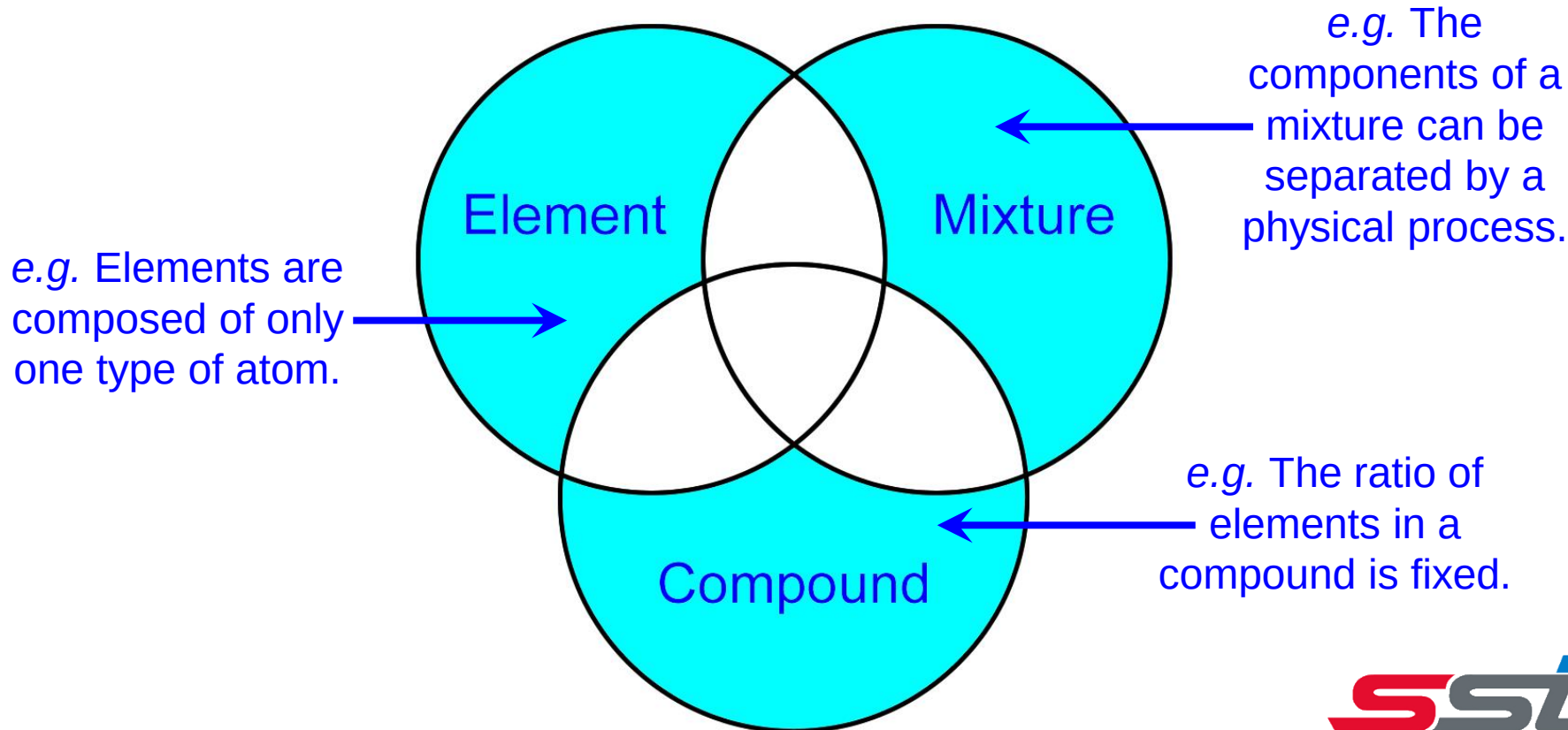
In what ways are the *pairs similar* to each other?



Higher Order Thinking Skills

- Roberts, J. L., & Inman, T. F. (2007). *Strategies for differentiating instruction*. Waco, Texas: Prufrock Press, Inc.

In what ways are elements, compounds and mixtures *different* from each other?



THE MILK BOTTLE EXPERIMENT

1. Predict what will happen.
2. Observe what happens.
3. Explain what has happened.



THE MILK BOTTLE EXPERIMENT

1. Predict what will happen.
2. Observe what happens.
3. Explain what has happened.



THE MILK BOTTLE EXPERIMENT

1. Predict what will happen.
2. Observe what happens.
3. Explain what has happened.



- Individual thinking.
- Group discussion (brainstorming and evaluation).
 - PowerPoint / Keynote design.
- Virtual gallery walk.
- Teacher facilitated class discussion.

APPLIED LEARNING AT SST



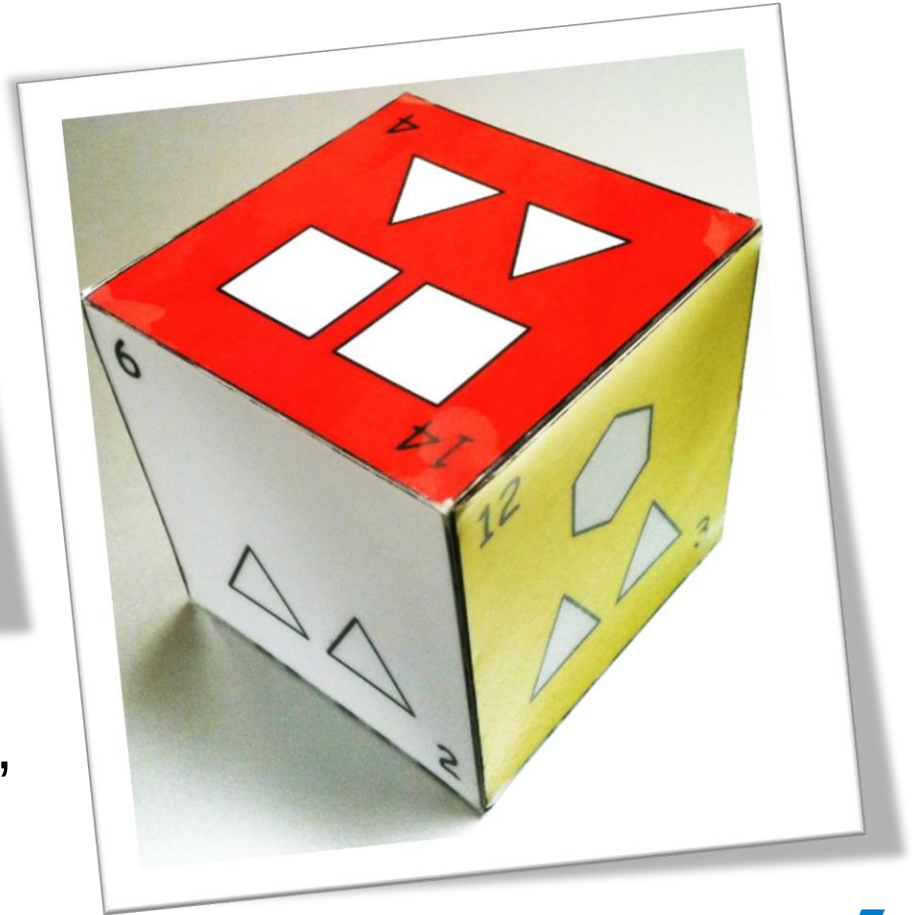
Mystery Tubes

- Each cardboard tube has four holes, through which there is a piece of string.
- Students gently pull the pieces of string and observe what happens.
 - Students suggest possible relationships between the variables.
- Students draw conclusions about what cannot be observed directly.
 - The activity challenges the students to think critically.
- The activity can be used as an introduction to atomic structure.

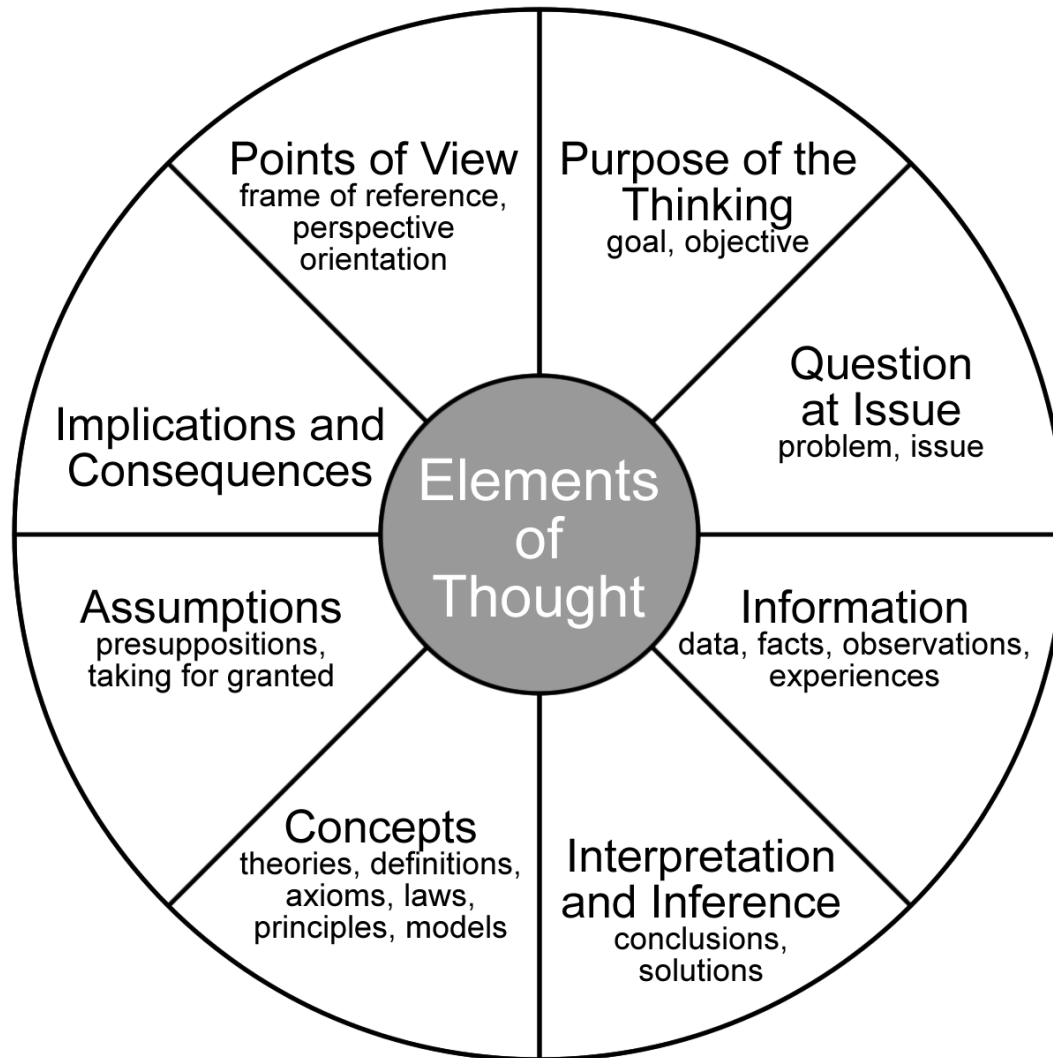
Curiosity Cubes



- From the information given, deduce what the pattern on the bottom of the cube is.

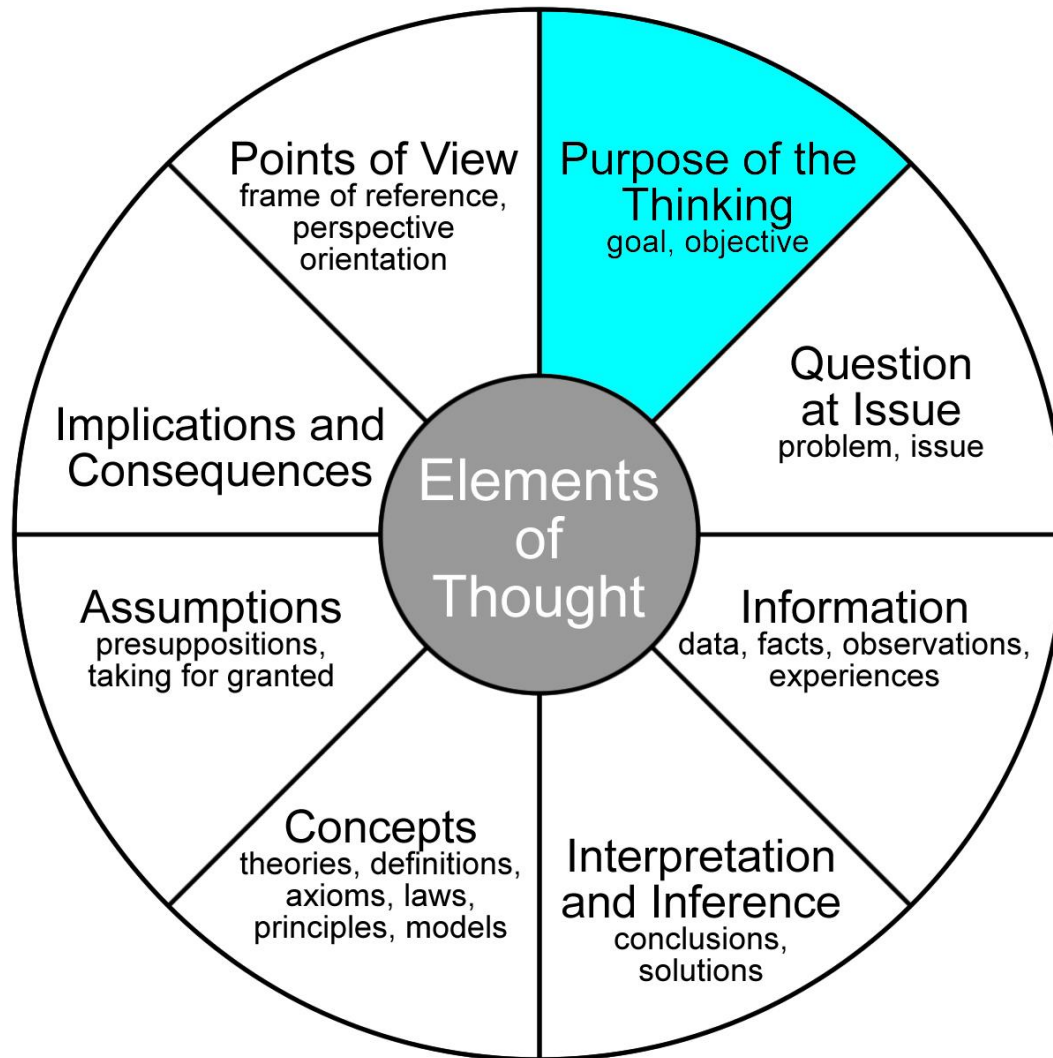


Paul's Wheel of Reason for Experimental Design



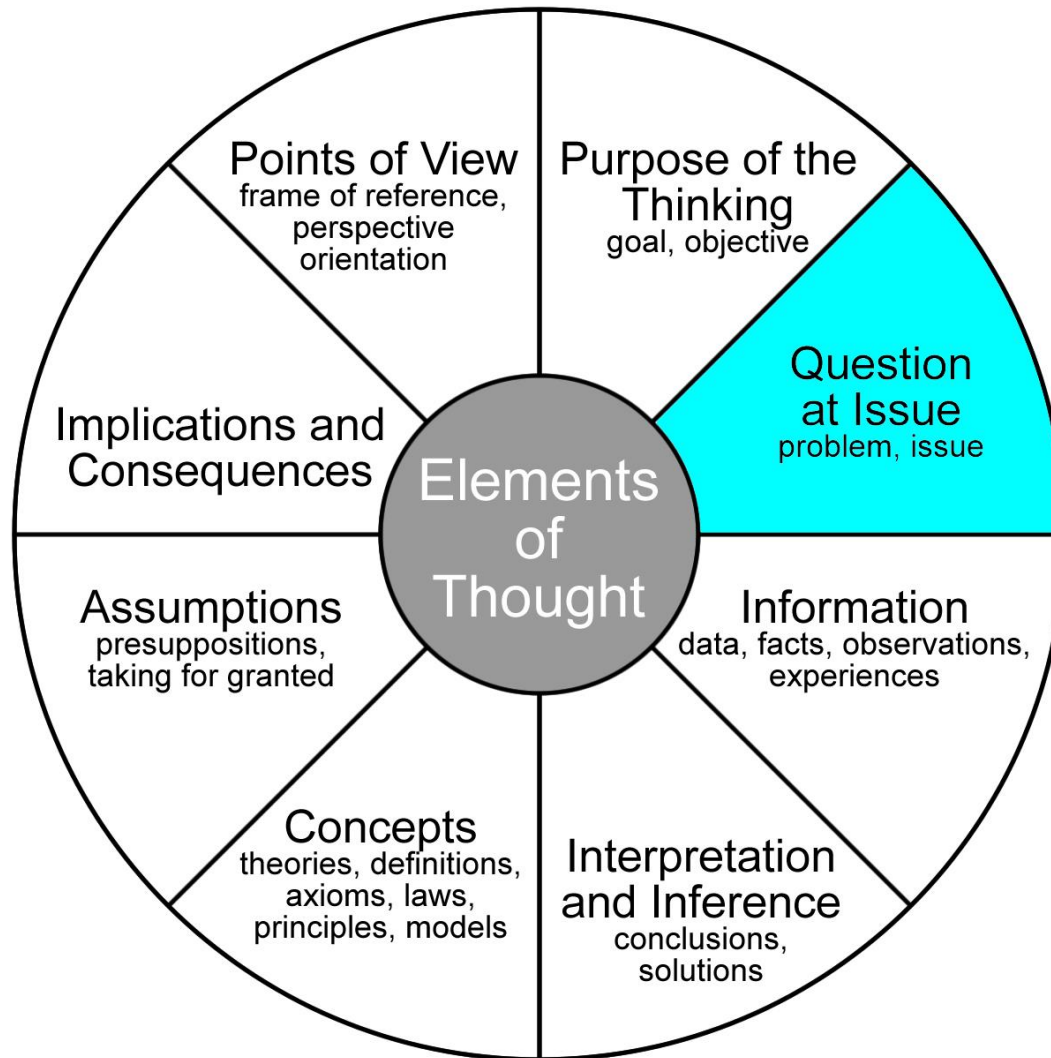
- Experiments should be designed using critical thinking skills.

Paul's Wheel of Reason for Experimental Design



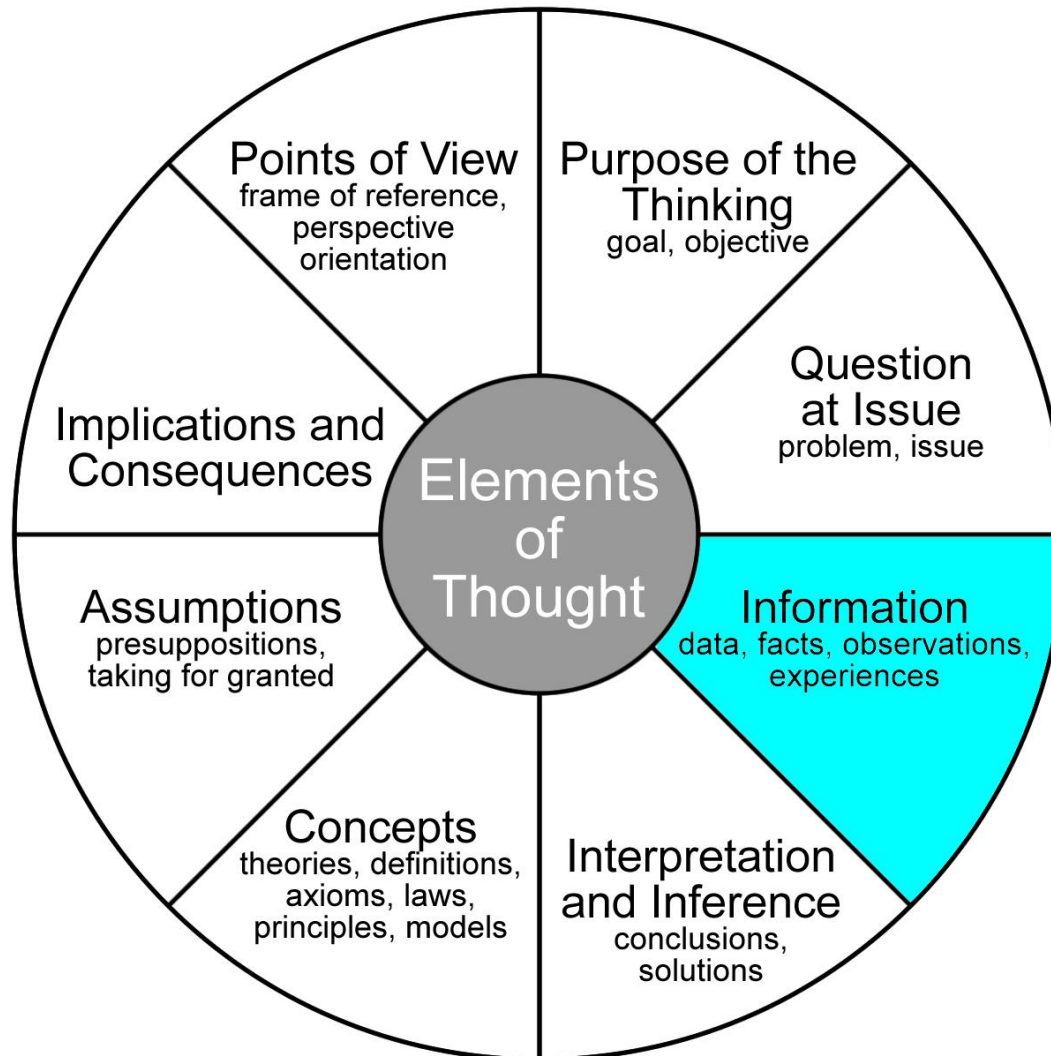
- What is the reason for performing the experiment?
- What hypothesis is being proved / disproved by performing the experiment?

Paul's Wheel of Reason for Experimental Design



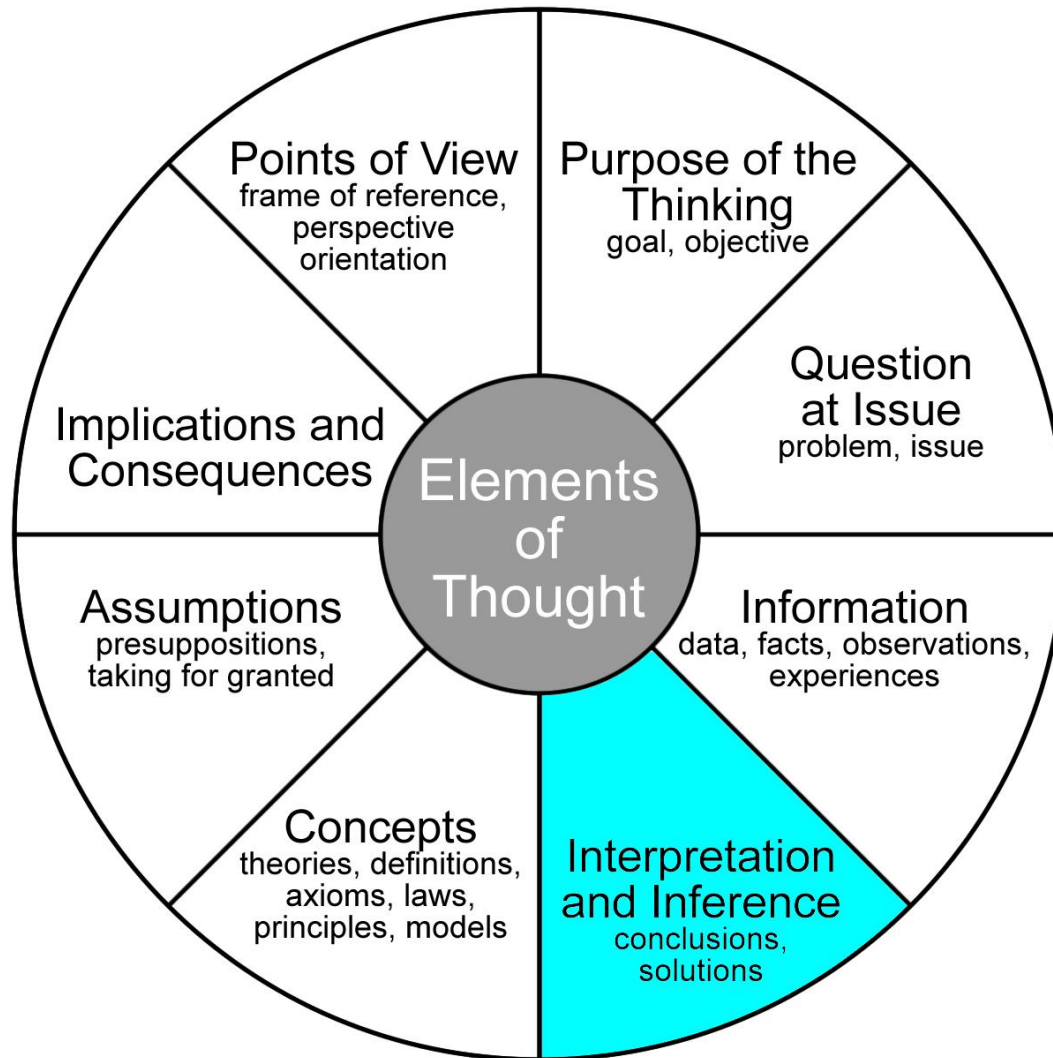
- Clearly state exactly what the experiment is being designed to do.
- What is the problem statement for the experiment?

Paul's Wheel of Reason for Experimental Design



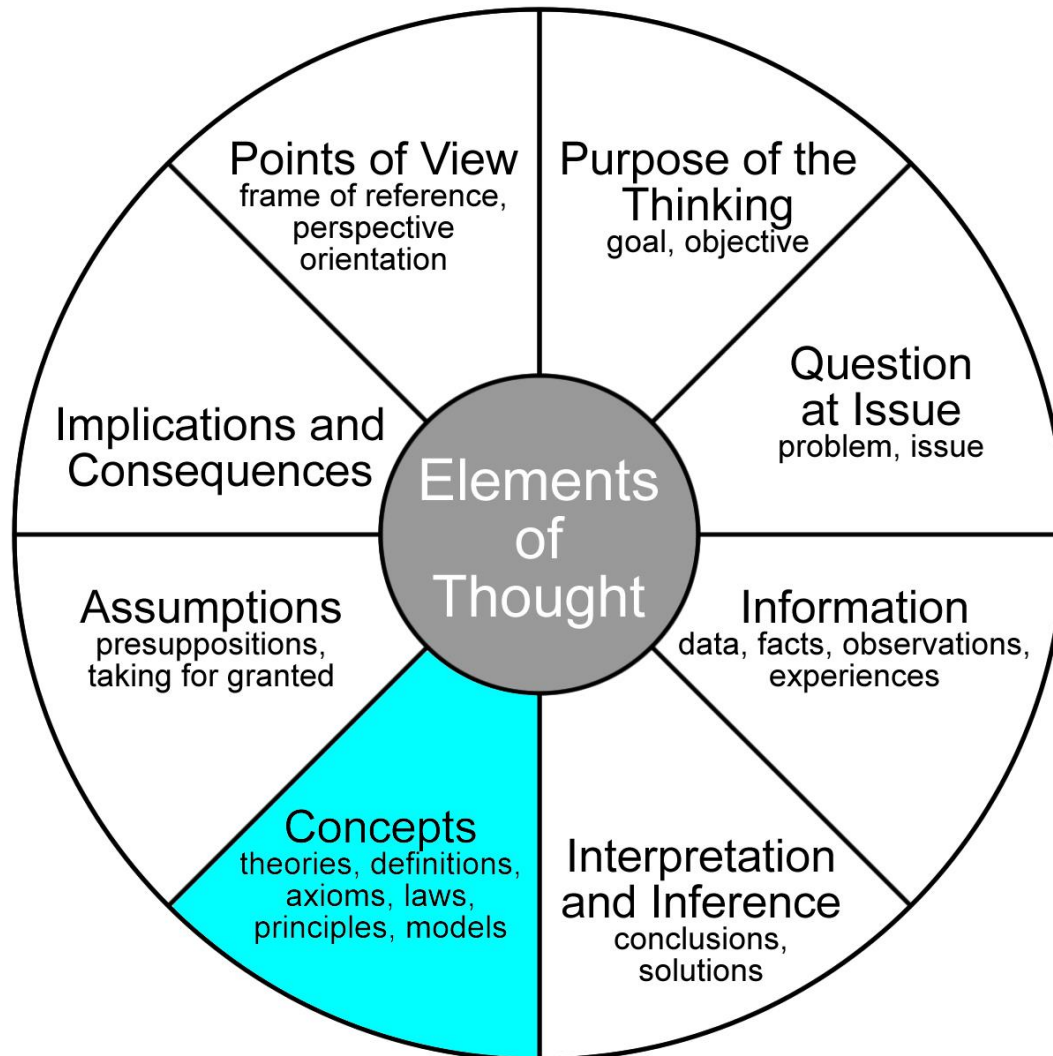
- What essential information needs to be collected by the experiment?
- What observations need to be made?
- What are the different variables (independent, dependent and control)?

Paul's Wheel of Reason for Experimental Design



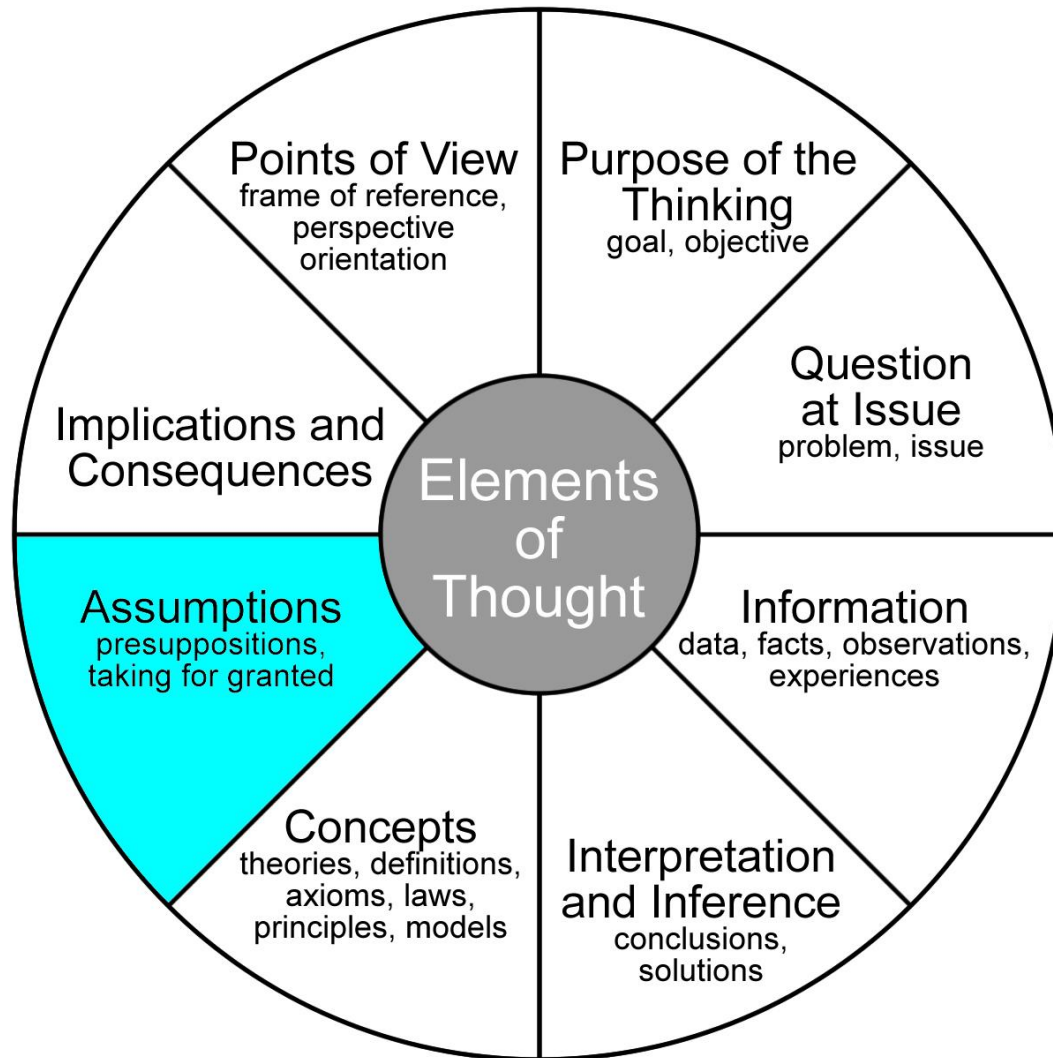
- How must the data from the experiment be manipulated and presented?
 - What calculations (if any) need to be done?
- What graphs (if any) need to be plotted?

Paul's Wheel of Reason for Experimental Design



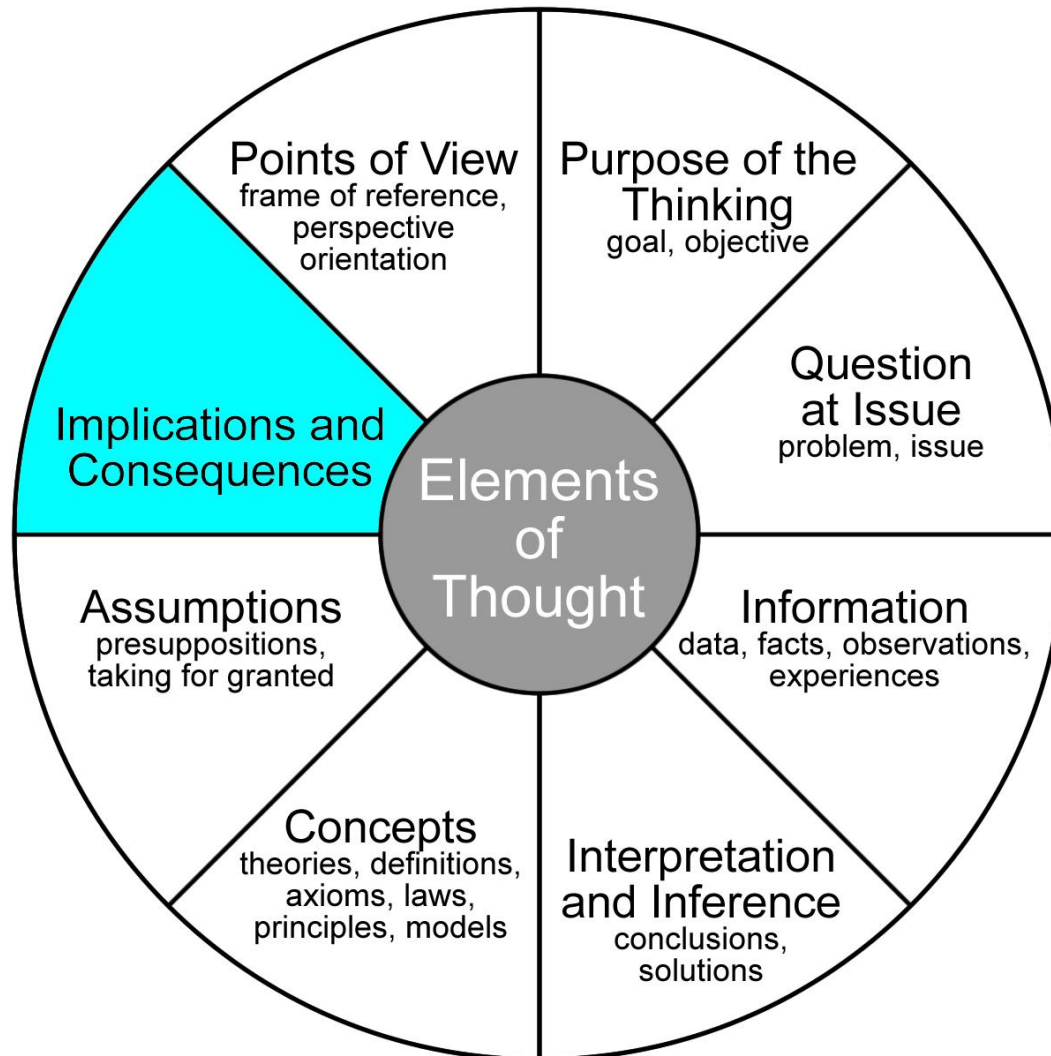
- What essential concept / law does the design of the experiment rely on? For example, the concept of energy and the law that energy can not be created or destroyed, but only converted from one form into another.

Paul's Wheel of Reason for Experimental Design



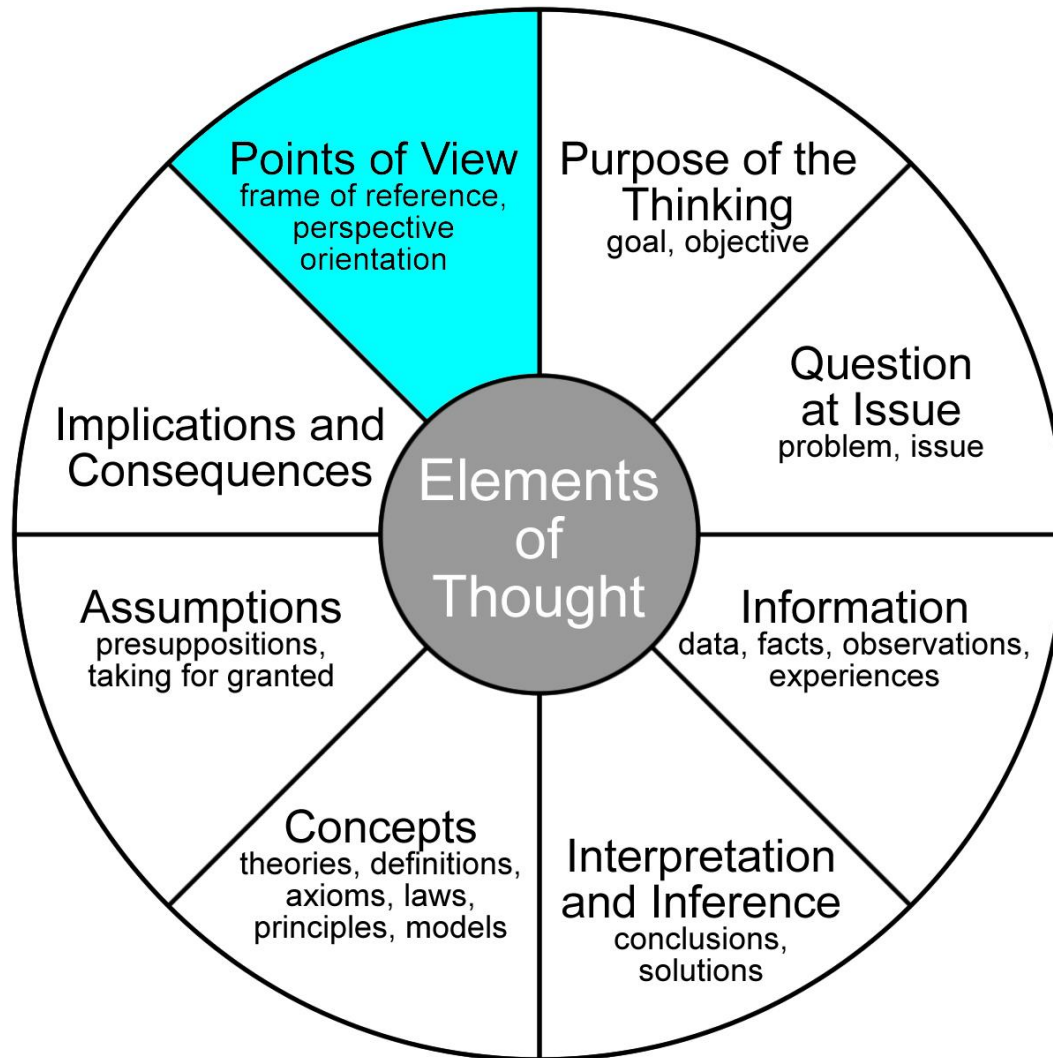
- What assumptions are being made when designing the experiment?
- Is it reasonable to make these assumptions?
 - Identifying assumptions can help to identify errors.

Paul's Wheel of Reason for Experimental Design



- What are the implications and consequences if the results obtained from the experiment prove / disprove a specific hypothesis?

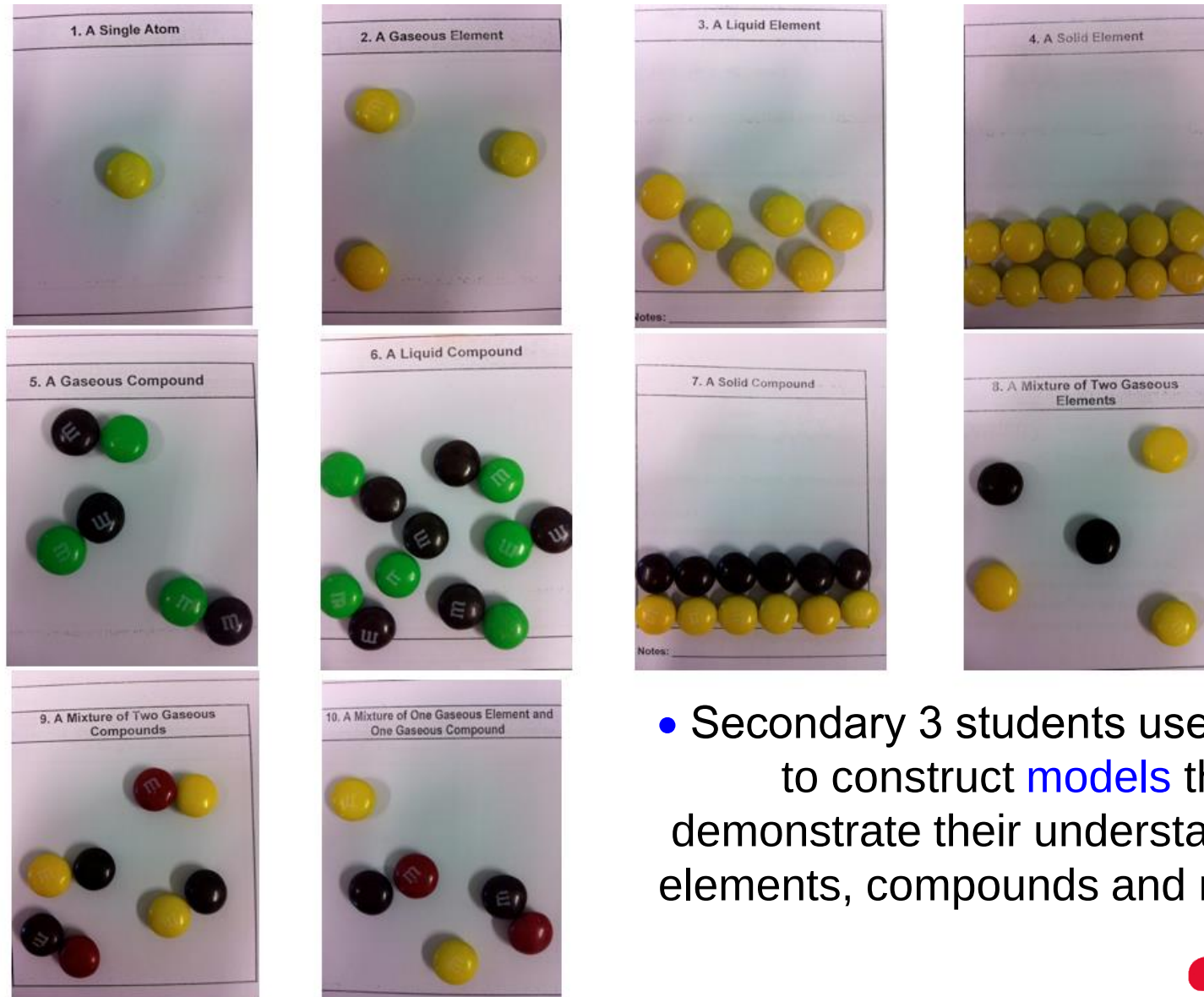
Paul's Wheel of Reason for Experimental Design



- Is there more than just one way of doing this experiment?
- If there is more than one way, which way is the most valid and reliable?

INFORMATION **COMMUNICATION** **TECHNOLOGY**

APPLIED LEARNING WITH ICT



- Secondary 3 students use M&M's to construct **models** that demonstrate their understanding of elements, compounds and mixtures.

APPLIED LEARNING WITH ICT

Concept Based Teaching

Developing the Concept of Organic Compound

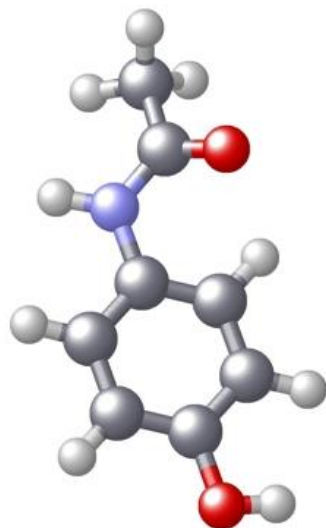


What are some examples of *organic compounds* and *inorganic compounds*?

APPLIED LEARNING WITH ICT

Concept Based Teaching

Developing the Concept of Organic Compound



- Students take photographs of materials that they think are organic / inorganic. The photographs are uploaded into a shared document.



APPLIED LEARNING WITH ICT

Concept Based Teaching

Developing the Concept of Organic Compound

Organic



In-organic



ORGANIC

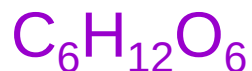
CO₂

EXTINGUISHED

Concept Based Teaching

Developing the Concept of Organic Compound

Organic



Inorganic



APPLIED LEARNING WITH ICT

- Alkanes are *hydrocarbons* with the general formula C_nH_{2n+2} .
- Alkanes react with oxygen to produce carbon dioxide and water.
 - Design a *spreadsheet* that will, when the number of carbon atoms in the molecule is keyed in, automatically calculate the energy change for the combustion of one mole of an alkane.

Enthalpy Change of Combustion Spreadsheet for Alkanes - Microsoft Excel non-commercial use

Spreadsheet to Calculate the Enthalpy Change of Combustion for One Mole of an Alkane

Definition: The enthalpy change of combustion is the **energy change** produced when **one mole** of a chemical **completely** reacts with **oxygen** under **standard conditions**.

Instructions:

- Choose an alkane by typing a whole number into the cell that is highlighted in blue and press "Return" on your keyboard.
- The spreadsheet will generate the balanced chemical equation for the complete combustion of one mole of the alkane.
- Using average bond energies, the spreadsheet will calculate the enthalpy change of combustion for one mole of the alkane.

Average Bond Energies:

C-C = 348 kJ mol⁻¹ C-H = 412 kJ mol⁻¹ O=O = 496 kJ mol⁻¹
 C=O = 743 kJ mol⁻¹ O-H = 463 kJ mol⁻¹

Balanced Chemical Equation:

C₄ + H₁₀ + 2 O_{2(g)} → 1 CO_{2(g)} + 2 H_{2O(l)}

Names of the First Ten Straight Chain Alkanes

CH₄ = Methane
 C₂H₆ = Ethane
 C₃H₈ = Propane
 C₄H₁₀ = Butane
 C₅H₁₂ = Pentane
 C₆H₁₄ = Hexane
 C₇H₁₆ = Heptane
 C₈H₁₈ = Octane
 C₉H₂₀ = Nonane
 C₁₀H₂₂ = Decane

Energy Change During Covalent Bond Breaking:

0	x	C-C	=	0	x	+348	=	0
4	x	C-H	=	4	x	+412	=	+1648
2	x	O=O	=	2	x	+496	=	+992
Total =							+2640	

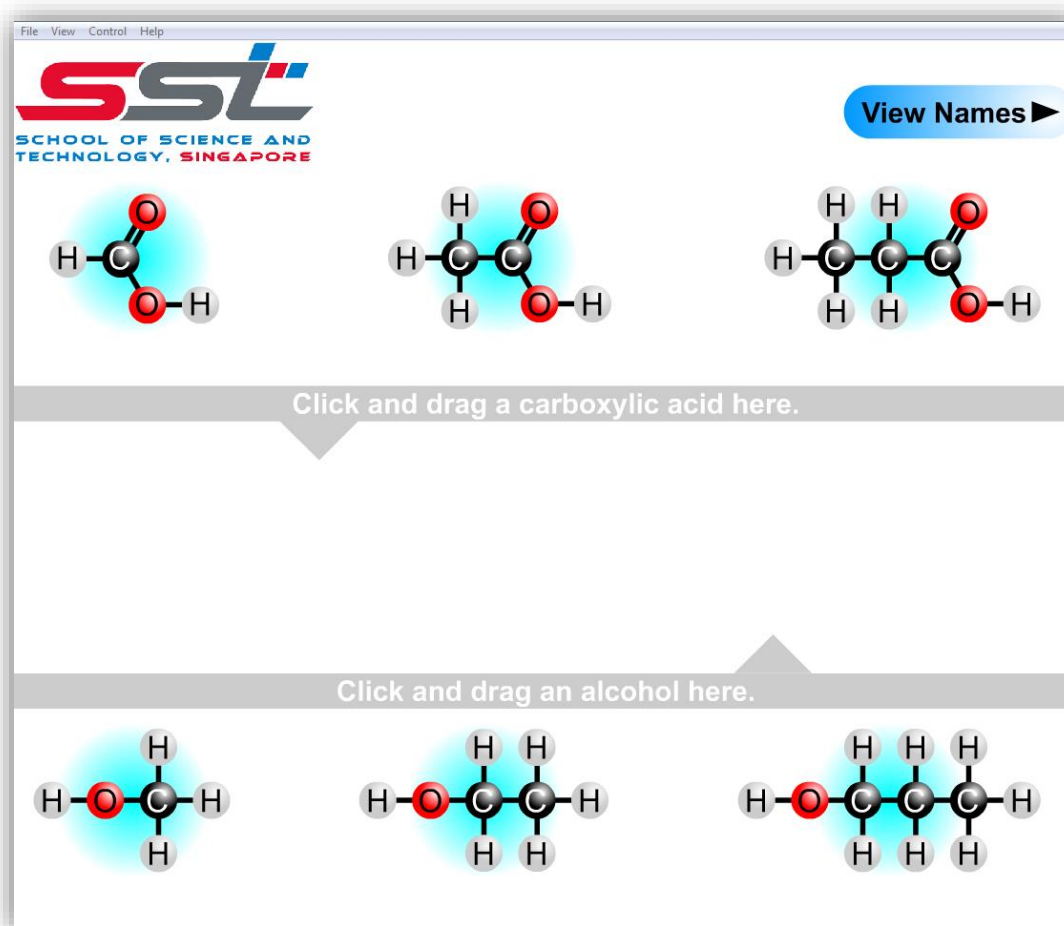
Energy Change During Covalent Bond Formation:

2	x	C=O	=	2	x	-743	=	-1486
4	x	O-H	=	4	x	-463	=	-1852
Total =							-3338	

Overall Energy Change for the Reaction:

=	+2640	+	-3338
=	-698	kJ mol ⁻¹	

APPLIED LEARNING WITH ICT



- Interactive animations to help the students visualise and understand difficult concepts.

www.chemical.sg

Quiz on Macromolecules

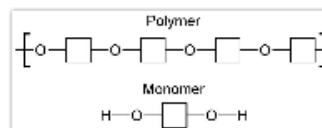
Question 4 of 15 \ Multiple Choice

A section of polymer, and the monomer that is used to make it, are shown.

Note: Click on the diagram to enlarge it.

The monomer undergoes condensation polymerisation. What is made each time a monomer adds to the polymer?

- ☐ A) Hydrogen molecules.
- ☐ B) Hydroxide ions.
- ☐ C) Oxygen atoms.
- ☐ D) Water molecules.



Submit

- Online platforms for learning and assessment.

APPLIED LEARNING WITH ICT

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Choose exactly what you want to learn!

You can filter to exclude exercises requiring a calculator or focus only on the hard questions. Or the easy ones if you are in the middle of a noisy coffee shop.

Choose the way you want to study!

You can choose to gradually move from topic to topic, get questions from all parts of the material you selected mixed together or choose the exam mode to really see how you are doing.

Need a break?

Return to learning sessions whenever you want to. You can come back to studying large topics without having to start all over again. And you can squeeze in some study time into gaps in your busy schedule.

Know yourself!

See detailed statistics on your performance by sub-topic so that you can focus on your weak spots. Nothing beats knowing how you are doing now for putting together your exam preparation strategy.

Name	Progress	Status	Action
Test case assignment 1	100%	Completed	View
Test case assignment 2	100%	Completed	View
Test case assignment 3	100%	Completed	View
Test case assignment 4	100%	Completed	View
Test case assignment 5	100%	Completed	View
Test case assignment 6	100%	Completed	View
Test case assignment 7	100%	Completed	View
Test case assignment 8	100%	Completed	View
Test case assignment 9	100%	Completed	View
Test case assignment 10	100%	Completed	View
Test case assignment 11	100%	Completed	View
Test case assignment 12	100%	Completed	View
Test case assignment 13	100%	Completed	View
Test case assignment 14	100%	Completed	View
Test case assignment 15	100%	Completed	View
Test case assignment 16	100%	Completed	View
Test case assignment 17	100%	Completed	View
Test case assignment 18	100%	Completed	View
Test case assignment 19	100%	Completed	View
Test case assignment 20	100%	Completed	View

- Online platforms for learning and assessment.
www.toktol.com

WHAT
QUESTIONS
DO YOU HAVE?

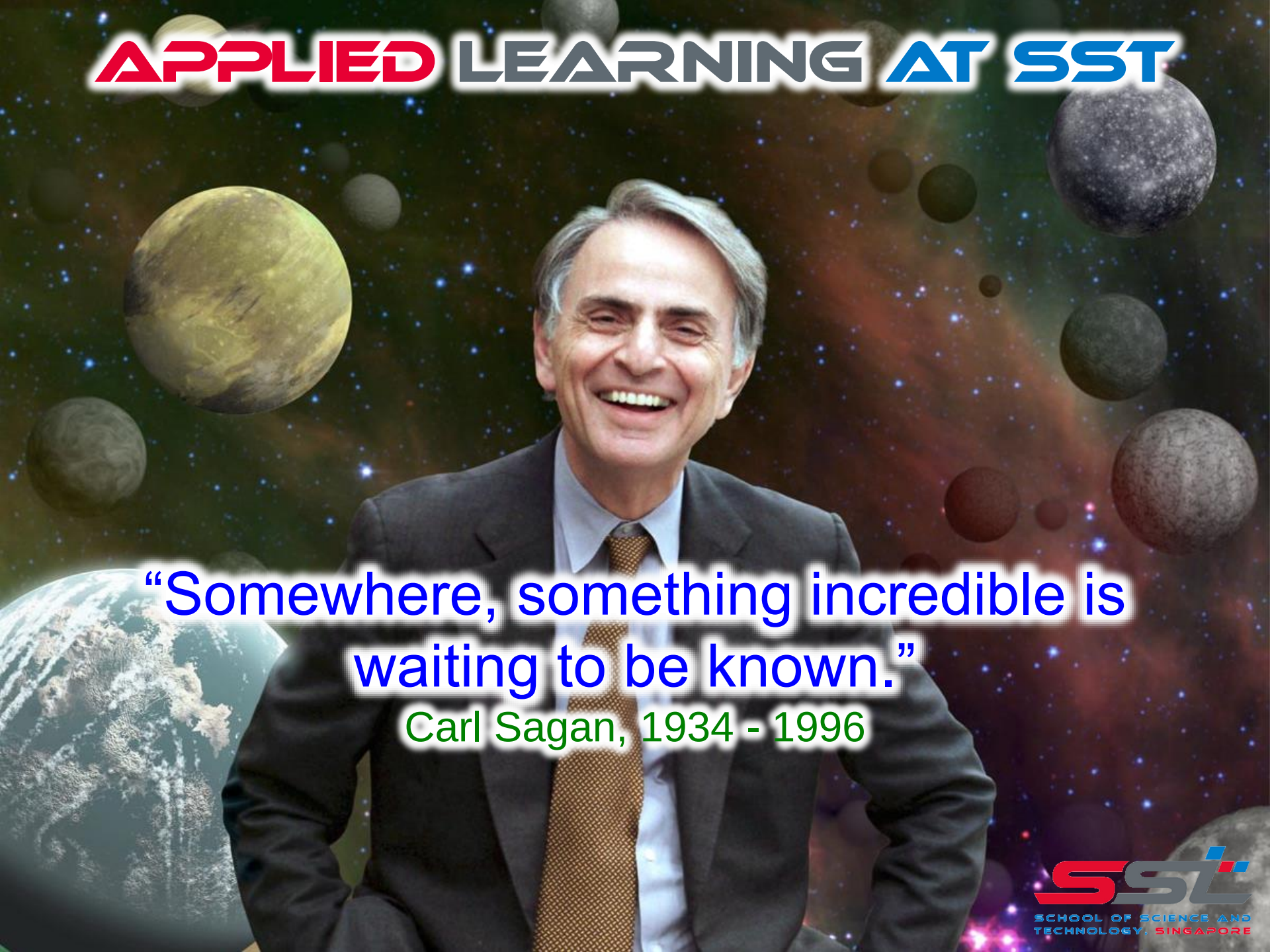
APPLIED LEARNING AT SST



“We live in a society exquisitely dependent on science and technology, in which hardly anyone knows anything about science and technology.”

Carl Sagan, 1934 - 1996

APPLIED LEARNING AT SST



“Somewhere, something incredible is
waiting to be known.”

Carl Sagan, 1934 - 1996

APPLIED LEARNING AT SST



“To bake the perfect apple pie from scratch, you must first invent the universe.”

Carl Sagan, 1934 - 1996

Presentation on:
Making Meaning out of Learning
The Rise of Applied Learning

For the:
Redesigning Pedagogy
International Conference – 2015

by Dr. Chris Slatter
christopher_slatter@sst.edu.sg

School of Science and Technology
(Singapore)
1 Technology Drive
Singapore
138572